

# Graphene Field-Effect Transistors for Sensing Ion-Channel Coupled Receptors: Toward Biohybrid Nanoelectronics for Chemical Detection

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Graphene field effect transistors (G-FETs) have appeared as suitable candidates for sensing charges and have thus attracted large interest for ion and chemical detections. In particular, their high sensitivity, chemical robustness, transparency, and bendability offer a unique combination for interfacing living and soft matters. Here demonstrated their ability to sense targeted biomolecules is demonstrated, by combining them with ion channel-coupled receptors (ICCRs). These receptors are naturally or artificially expressed within living cell membranes to generate ion fluxes in the presence of chemicals of interest. Here, those biosensors are successfully combined with a G-FET array which converts the bio-activation of the ICCRs into readable electronic signals. This hybrid bioelectronic device leverages the advantages of the biological receptor and the graphene field effect transistor enabling the selective detection of biomolecules, which is a current shortcoming of electronic sensors. Additionally, the G-FET allows for discrimination of the polarity of the ion fluxes which otherwise remains hidden from conventional electrophysiological recordings. The multisite recording ability offered by the G-FET array raises numerous possibilities for multiscale sensing and high throughput screening of cellular solutions or analytes, which is of both fundamental and applied interest in health and environment monitoring.

## 1. Introduction

There are several approaches to measuring the content of analytes in liquids (e.g., spectroscopy, chromatography, mass spectrometry, electrochemical methods, immunoassays, and molecular biology techniques). The choice of method depends on factors such as the nature of the analyte, the required sensitivity, the sample matrix, the available instrumentation, and the operation conditions. Among them, extracellular devices such as microelectrodes or field effect transistors enable noninvasive, long-lasting, real-time, and on-site monitoring. They are compact, easily integrated into portable or wireless systems, and compatible with microfluidic and large-scale implementation. More recently, several successful demonstrations have combined them with promising biological sensors, comprising cells, receptors, peptides, and nanovesicles that have been recently reviewed.<sup>[1]</sup> However, measuring ion channels within living

cells has long been a limit of detection for bioelectronic devices, mainly because of their size (nm) and the low currents (pA) involved. Recently, emerging graphene devices have shown early signs that the technology is becoming mature enough to surpass this limit.<sup>[2,3]</sup> Because of the broad range of ligand-gated ion channels, their integration with electronic sensors is of particular interest to identify and enlarge the number of detectable chemicals. In particular, ion channels (ICs) generate a fast and large flow of ions ( $\mu\text{s}$  to  $\text{ms}$ , and up to millions per second respectively) that might be sensed by ion-sensitive field effect transistors (IS-FETs) and converted into readable electrical signals. This versatile tool can be easily adapted to various ISFET-ICCR couples, allowing the detection of small amounts of biochemical compounds in highly concentrated analytes. Moreover, ISFETs are easily reproduced on multisite arrays so that biological objects of various structures (e.g., ion channels, large cells, cell networks) can be probed in real-time, on different sites.

Currently, the number of endogenous ligands recognized by vertebrate ligand-gated ion channels is very small ( $<10$ ),<sup>[4]</sup> which limits their applications in the field of biosensing. To expand the

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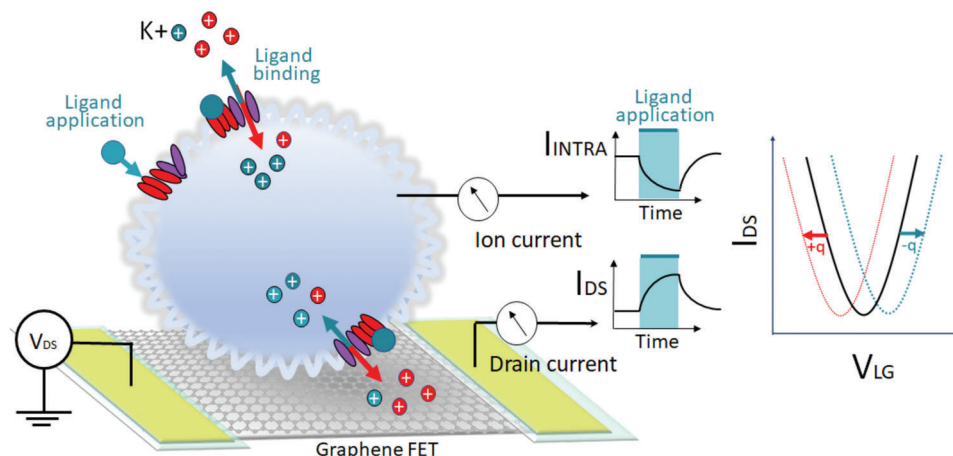
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**Figure 1.** Principle of hybrid bioelectronic sensing. The schematics describe a graphene field effect transistor interfacing a *Xenopus* oocyte heterologously expressing engineered ion channels. The chemicals to be sensed (teal disk, labeled ligand) bind to the receptor part (red) and trigger the opening of the transmembrane ion channel (purple), which results in a change of the ion flow through the channels. Following the ligand-activation of ICs, the charges in excess ( $K^+$  mainly, blue + circles) above the graphene transistors channel can be detected by monitoring the G-FET drain current. On the right, is the expected shift of the G-FET transfer curve ( $I_D$ – $V_{LG}$  curve) for both accumulation (+Q) or depletion (–Q) of positive ions. The G-FET dimensions are not at scale. Indeed, the graphene surface is  $\approx 10^4$  times smaller than the oocyte surface. Therefore, the graphene sensor is extended to a membrane surface including  $\approx 10$  ICs.

repertoire of recognized ligands, two approaches have been explored in the present study with non-vertebrate ICs and artificial ICs with exogenous receptor moieties. In particular, a highly conserved odorant receptor co-receptor (Orco) from *Drosophila melanogaster* (fruit fly, a common scientific model), generates high current amplitudes in response to a few synthetic ligands in *Xenopus* oocytes,<sup>[5]</sup> and it was consequently used as a first model in this study. On the other hand, artificial ICs have been created by fusing ICs with G protein-coupled receptors (GPCRs),<sup>[6,7]</sup> which are called ICCRs. Integrated in cell membranes, ICCRs are suitable for chemical sensing operation in a physiological fluid, taking advantage of specific combinations of ligand–receptor for real-time detection and identification of various biomarkers or chemical compounds in general. When the molecule to be sensed (ligand) binds to the protein receptor, conformational changes of the protein trigger the opening or closing<sup>[8]</sup> of ICs that are coupled, as illustrated in **Figure 1**. A hindrance to this approach could be the inherent challenges of manipulating and interfacing living organisms. However, compared to artificial membranes, *Xenopus* oocytes offer specific advantages such as a natural lipid bilayer asymmetry and composition including specific lipids (e.g.,  $PIP_2$ , cholesterol) often critical for receptor and ion channel activities. Moreover, the oocytes preserve gradients of ions between intra- and extracellular compartments, enabling ligand-induced ion flow in the absence of imposed electrical forces such as membrane voltage-clamp. Because of such high gradients of ion concentration between intra and extracellular media, this modulates ion fluxes through the lipid membrane that can be detected by electrophysiological methods.

The initial demonstration of ICCR biosensing<sup>[2]</sup> was carried out on *Xenopus* oocytes which are a standard heterologous expression system for ICs. These living cells have several advantages for operation conditions in standard laboratories. Their optimal condition is around room temperature (19 °C) with no requirement for temperature control equipment. Additionally,

they do not require  $CO_2$  and are maintained for days in minimum salt buffers, and up to hours when peeled (without their external vitelline envelope). Furthermore, they remain quiescent throughout the experiments, exhibiting no cell division or change in morphology. The detection of ligand-induced currents is usually performed in whole-cell configuration with the two-electrode voltage-clamp (TEVC) method when the ligand binding sites are on the extracellular side of the receptors.<sup>[9]</sup> In this method, oocytes are impaled by two electrodes and the throughput level is low in manual setups. While TEVC robots have been developed in 96-well plate format since, the level of throughput could be largely increased, as well as avoiding invasive cell manipulation should significantly increase the observation time window. To make progress towards this objective, extracellular measurements have been developed that do not alter the cell membrane or intracellular fluid and enable long-lasting recording of cell activity over weeks.

One drawback is that extracellular variations of potential or ion concentration are lower than intracellular changes,<sup>[10]</sup> and thus more difficult to sense. As illustrated Figure S1 (Supporting Information), this results from the high impedance of the cell membrane, the leakage currents to the grounded bath, or the charges screening in the solution. The typical ion concentration of the bath is  $\approx 100$  mM, and the Debye length is significantly reduced (1 nm, Figure S2, Supporting Information) compared to ultrapure water. Moreover, the ICs of the oocytes are embedded in a membrane that is rather folded at the microscale such that ICs could be a few microns away from the device and at least above the Debye screening length (1 nm at 100 mM ionic strength of the electrolyte solution, Figure S2, Supporting Information). In addition, the number of detectable ICs is significantly reduced in comparison with whole-cell TEVC measurements, reducing the detection efficiency. These limitations might be overcome by overexpressing the ICs of interest to increase their density around the recording site. Also, the use of highly sensitive and

biocompatible materials such as graphene on sapphire or glass<sup>[11,12]</sup> is a way to promote the adhesion of the cell membrane to ensure high sealing resistance and high ISFET performance.

Due to their high sensitivity, affinity with the cell membrane, and chemical inertness, G-FETs have appeared as a promising candidate for such ion detection in physiological fluids. This 2D honeycomb lattice of carbon atoms exhibits high charge-carrier mobility and promotes the adhesion of cells, providing high sensitivity to extracellular potential changes (20  $\mu$ V) and wide bandwidth to sense both low and high-frequency biological signals (mHz–MHz) with minimal thermal or chemical drift. This is further supported by the chemical inertness of monolayer graphene, which enables to exposure of the transistor channel directly to liquids at the closest vicinity of the environment to be sensed, such as an ultrathin electrical double layer standing alone at the interface with the liquid or the cell to-be-sensed. This unique property of combining the high mobility of charge carriers and an atomically thin interfacial layer enables it to reach higher sensitivity and frequency regimes than its counterpart silicon or carbon nanotube FETs. The first demonstrations were achieved for sensing ion channels,<sup>[2]</sup> bioelectric signals,<sup>[13]</sup> and ions in solution.<sup>[14]</sup> Since the use of G-FETs has been extended for sensing analyte composition,<sup>[15,16]</sup> DNA,<sup>[17]</sup> pH,<sup>[18,19]</sup> virus,<sup>[20]</sup> and ionic signals generated by electrogenic cells such as neurons<sup>[21,22]</sup> or cardiomyocytes.<sup>[13,23]</sup> These examples illustrate the broad range of possible applications for graphene-based ISFET in the fields of health and environment. Enhanced performances such as high selectivity and sensitivity could be achieved by functionalizing graphene directly with bio-recognition elements, such as enzymes or antibodies<sup>[24]</sup> or ionophore membranes.<sup>[15]</sup> However, this limits the detection of the molecules where their bio-recognition elements exist and can be used to functionalize graphene. Also, this restricts the detection of a single and predefined species per G-FET. Recent attempts such as the introduction of the Hall or field effects and the use of pristine graphene<sup>[3,25]</sup> might overcome this limit.

In this study, bare graphene FET were directly exposed to programmable ICCR bringing promising outcomes 1) for real-time and extracellular sensing of ion channels at the microscale within living matter, and 2) for enlarging the available receptors among the more than 800 human members. The two being of fundamental and applied interests. The field effect detection of ICCRs by G-FET is described in Figure 1 and Figure S1 (Supporting Information). The ions adsorbed on the graphene induce an electrostatic gating. The surface potential is changed which tunes the Fermi level. As a result, the charge neutrality point  $V_D$  is shifted by a value, which depends on the ion concentration. Therefore, the ionic concentration of a solution can be deduced by monitoring the gate potential  $V_{LG}$  needed to reach the charge neutrality point. This technique is well adapted to detect the proton concentration of an electrolyte and to determine the pH of solutions, as reported in several seminal works.<sup>[18,19,26]</sup>

Considering a perfect single crystal of graphene, its small amount of dangling bonds should not favor ion adsorption, which is a primary key for ion sensing. However, it was shown a few years ago that intrinsic defects of the chemical vapor deposition (CVD)-grown graphene-monolayer (e.g., sp<sup>3</sup>-hybridized

carbon atoms) constitute sites of high chemical reactivity for enhanced sensitivity.<sup>[27,28]</sup> Those specific sites favor a charge transfer between the solution and the graphene which allows them to reach high sensitivities to detect adsorbed ions or molecules in liquids<sup>[18,27]</sup> and with biological membranes.<sup>[3]</sup>

Here, we show the application of graphene microtransistors for sensing ligand-activated IC activity within *Xenopus* oocytes, and their potential for implementing future generation of hybrid bioelectronic sensors. Two types of receptors have been used as a proof-of-concept: the GPCRs (later called ICCRs) and an insect co-receptor (Orco). The GPCRs recognize a wide range of triggers from photons to large proteins, including hormones, ions, neurotransmitters, peptides, and even bacterial toxins. They are interesting for in vitro diagnostics or drug screening as major pharmaceutical targets. Orco is one subunit of the insect odorant receptors and future applications are envisioned in the screening of odorant molecules (e.g., pheromones for mass trapping of pests) or repellents against mosquitoes. In general, all-natural or engineered ion channels recognizing a ligand of interest could be integrated into this technology.

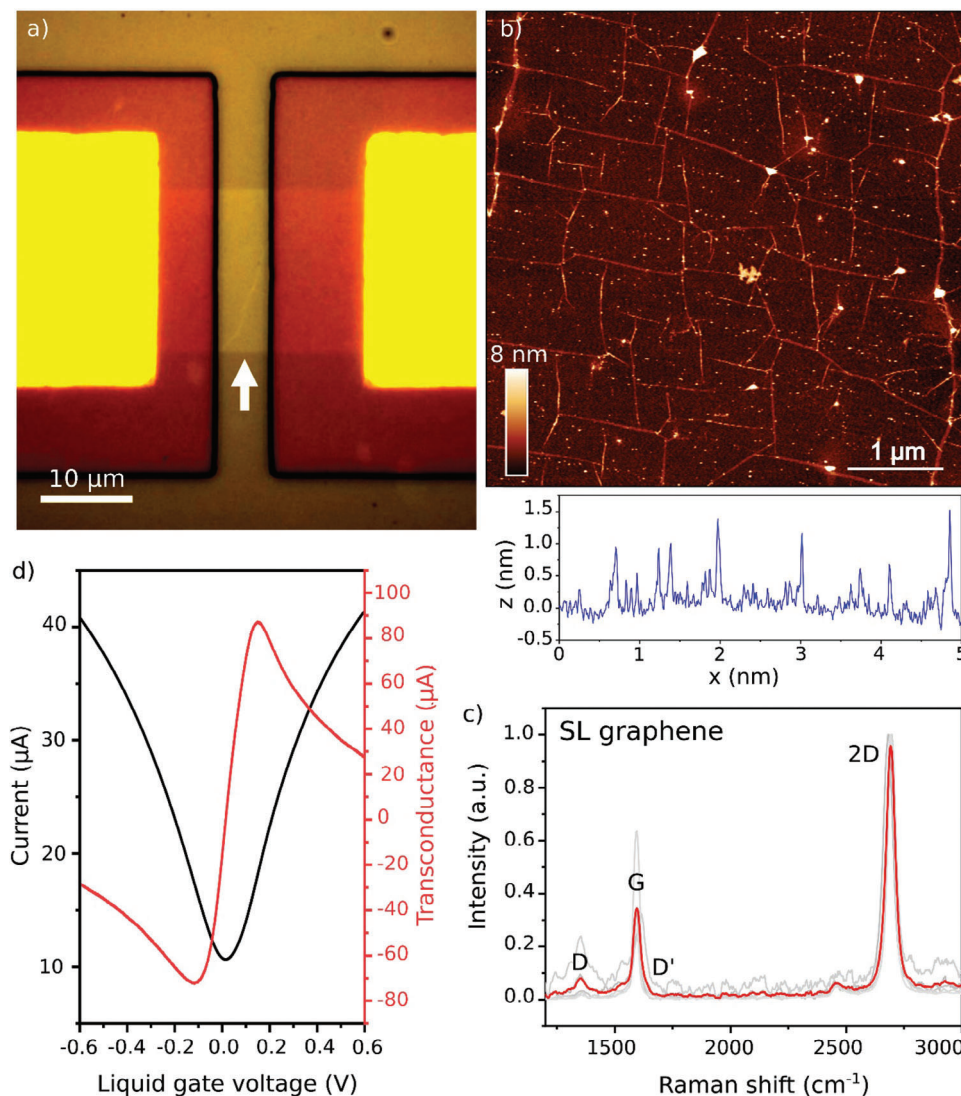
## 2. Results

### 2.1. Manufacturing Ion-Sensitive Graphene Field Effect Transistors

Because of its chemical stability graphene monolayer can be directly exposed to the cell and liquid media increasing considerably the device's performance in comparison with more conventional silicon or indium tin oxide materials. Indeed, the FET sensitivity is inversely proportional to the gate thickness, which is  $\approx 1$  and 10 nm for graphene and silicon FET respectively. Additionally, graphene nanomaterial can be reported on any kind of substrate including transparent ones, which provides optical access to the biological object of interest by using a standard transmission microscope. Here, the monolayer graphene was obtained by CVD and transferred from the copper growth substrate onto sapphire by wet transfer technique. Then arrays of graphene field-effect transistors have been fabricated in a clean room microfabrication facility (details in materials and methods).

Figure 2a shows a typical sample and zooms on a G-FET to illustrate the high quality of the resulting graphene channels directly exposed to the analytes. Only a few residuals of the fabrication process are observed, and nanoscale wrinkles can already be observed as highlighted by the arrows. The atomic force micrograph confirms the overall quality of the graphene transistor channel (Figure 2b). The monolayer is continuous with typical wrinkles of about a few nm height (section profile in blue) and a mean roughness of  $\approx 458$  pm which reveals the high quality of the monolayer graphene after the fabrication process. This value is below the expected Debye length (1 nm at 100 mM, Figure S2, Supporting Information) which demonstrates the high quality of the fabricated G-FET and their potential for ion sensing.

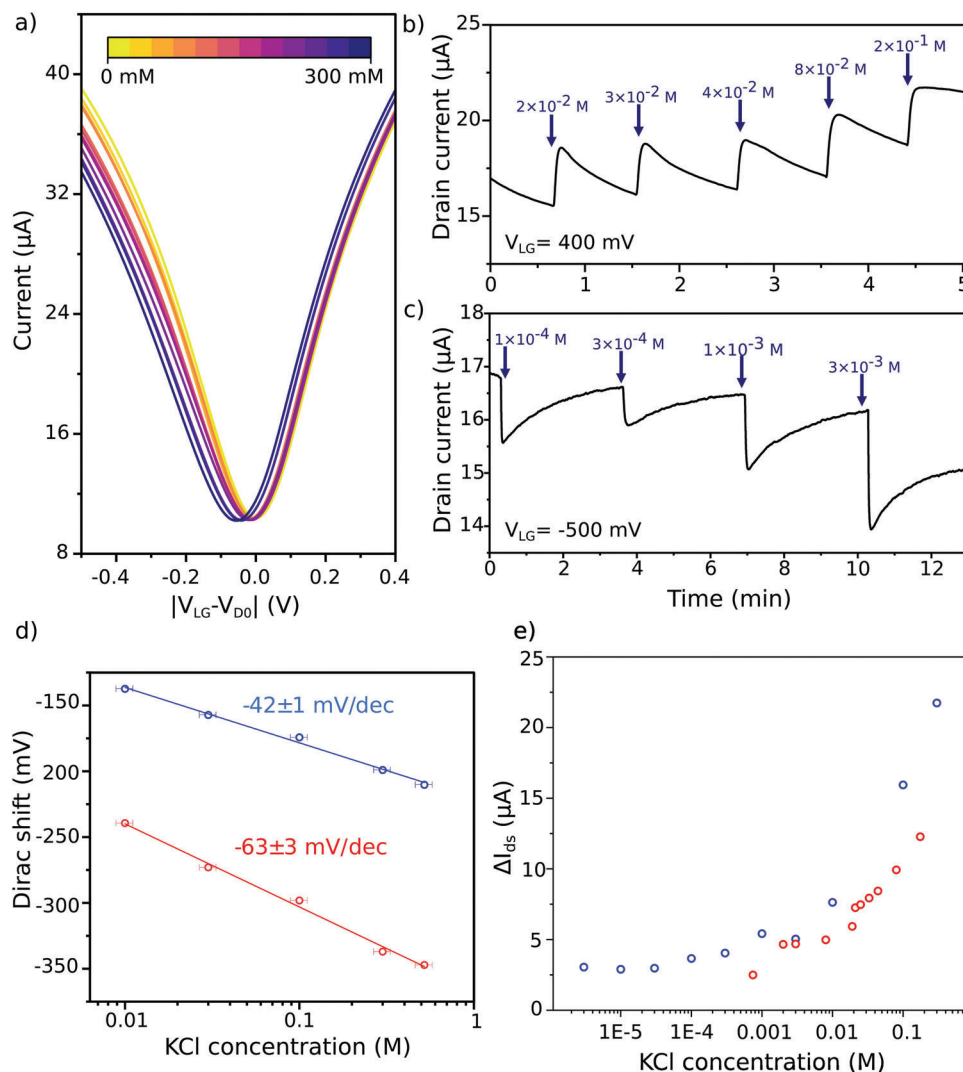
Raman spectra have been acquired on the pristine graphene and on the final sample to assess its quality in terms of contamination and defects (Figure 2c). Single spectra were acquired on different regions of the graphene channel using a laser excitation of 532 nm (2.33 eV) and a low incident power of 1 mW. Each spectrum results from 10 accumulations of 5 s acquisition. Two



**Figure 2.** Manufacturing graphene field-effect transistors. a) Optical micrograph of a  $20 \times 5 \mu\text{m}^2$  G-FET after the fabrication process. An arrow indicates the position of the graphene transistor channel which is directly exposed to the analyte or bio-object to be sensed. The drain and source metallic leads (brightness areas) are encapsulated with an insulating resist for liquid operations. b) Atomic force micrograph of a G-FET channel after passivation. A cross-section exhibits the characteristic height of wrinkles and impurities (1–5 nm). c) Raman spectroscopy averaged (red curve) on five spatial points (grey curves) within a G-FET channel. d) Transfer characteristic and transconductance of a G-FET measured in PBS.

main peaks are observed at  $1591 \pm 4$  and  $2693 \pm 2 \text{ cm}^{-1}$  corresponding to the G and 2D bands related with graphene. The intensity ratio  $I_G/I_{2D} = 0.27$ , the G-band peak frequency, and the full width at half maximum of the 2D peaks ( $\Delta\omega = 46 \text{ cm}^{-1}$ ) are as expected for monolayer graphene.<sup>[29,30]</sup> A single peak in the 2D band stands for a monolayer graphene. Moreover, the D-band peaks ( $1348 \text{ cm}^{-1}$ ) reveal the presence of defects within the pristine graphene and the G-FET channel. Comparing it with the D' peak ( $1626 \pm 3 \text{ cm}^{-1}$ ), which appears as a small shoulder of the G peak, enables us to probe the nature of the defect,<sup>[31]</sup> being mainly vacancy, impurity, and topological defects. In particular, the intensity ratio  $I_D/I_{D'} = 1.2$  indicates a majority of sp<sup>2</sup>-hybridized boundary defects that can be found within grain boundaries for polycrystalline CVD-grown graphene. Those defects are expected to support the field effect detection for sensing ions.<sup>[3,18,27]</sup>

Field-effect measurements have been performed to assess the G-FET sensitivity to potential variations in liquids (Figure 2d). The drain current  $I_D$  was monitored while varying the liquid gate potential  $V_{LG}$  from  $-600$  to  $600$  mV with a quasi-reference platinum electrode immersed in phosphate-buffered saline (PBS) solution (10 mM phosphate, 137 mM NaCl, 2.7 mM KCl, pH 7.4), pH = 7.4). The  $I_D$  versus liquid gate voltage  $V_{LG}$  reveals the expected ambipolar field-effect behavior of G-FET, with a slight p-type behavior at zero gate voltage, which was previously confirmed by Raman micro-spectroscopy for pristine CVD-grown monolayer.<sup>[11]</sup> The derivative of the liquid gate transfer curve enables to extraction of the transconductance  $g_m$  (red curve, Figure 2d) which is directly related to the G-FET sensitivity  $S = g_m/V_{DS}$  and enables to set of the proper operation setpoint. Here, the maximum sensitivity is  $\approx 1 \text{ mS V}^{-1}$  (at  $V_{LG} \approx 150$  mV).



**Figure 3.** The ion detection performance of the graphene field-effect transistors. a)  $I_D$ - $V_{LG}$  transfer curves of a  $20 \times 10 \mu\text{m}^2$  G-FET as a function of the ionic concentration of the liquid gate, at fixed drain-source voltage ( $V_{DS} = 80 \text{ mV}$ ). The liquid gate voltage  $V_{LG}$  is applied with a quasi-Pt reference electrode immersed in the solution. The concentration of KCl is increasing from 0 to 300 mM (yellow to blue respectively), leading to a shift in the position of the Dirac point toward negative values. b) and c) Evolution of the drain current  $I_D$  over time while increasing KCl concentration. In terms of injected charges, the value increases from  $1.10^{-6}$  to  $3.10^{-5}$  mol (upper trace) and from  $2.10^{-8}$  to  $7.10^{-6}$  mol (bottom trace). The electrolyte potential is fixed at (b)  $V_{LG} = 400 \text{ mV}$  and (c)  $-500 \text{ mV}$ . d) Dirac shift  $\Delta V_D = V_D - V_{D|C=0}$  corresponds to the value of the Dirac point compared to the value obtained at  $[C_{\text{ion}}] = 0 \text{ M}$  as a function of the KCl concentration, for two different G-FETs (blue and red). The slope corresponds to the G-FET sensitivity to variations in ionic concentration. (e) Drain current variations  $\Delta I_{DS} = I_{DS} - I_{DS|C=0}$  compared to the initial drain current at  $[C_{\text{ion}}] = 0 \text{ M}$  as a function of the KCl concentration. Blue points are extracted from the drain current value at fixed  $V_{LG} = 400 \text{ mV}$  on the transfer curves (from Figure 3a). Red points correspond to the real-time stabilized drain current value after KCl injection (from Figure 3b).

## 2.2. Ion Sensing with the Graphene FET

Beyond their interest in the present study, which targets the detection of inward rectifier potassium channels (Kir), potassium channels are also expressed in many cells, which makes its detection relevant for other applications. To that end, the transfer curves  $I_D$ - $V_{LG}$  were measured in KCl electrolytes and for several ionic concentrations. The G-FET conductance was monitored while varying the gate potential  $V_{LG}$  applied in the ionic solution whose volume and ion concentration were controlled. Starting from deionized water, a small amount of highly concentrated ion solution was added to the electrolyte to gradually in-

crease the ion concentration. Typically, we proceeded by successive injections of 1 M KCl solution. **Figure 3a** shows the evolution of the transfer curve as a function of several ionic concentrations, ranging from 0 to 300 mM (yellow to blue curves).

By increasing the KCl concentration, the charge neutrality point shifted towards negative values, indicating a negative doping of the graphene channel that could result from either chemical or electrostatic interactions. The adsorption or physisorption of  $\text{K}^+$  at the vicinity of the G-FET channel or the increased ionic strength of the electrolyte<sup>[14,32]</sup> can induce the accumulation of electrons within the transistor channel. The G-FET sensitivity for potassium detection is extracted from the dependence

of the Dirac shift  $\Delta V_D$  with the ionic concentration. The value is obtained using a linear fit on  $\Delta V_D(C_0)$  curves. The maximum value was  $-63 \pm 3$  mV decade<sup>-1</sup> (Figure 3d). This sensitivity with bare graphene is quite similar to what is obtained for membrane-coupled G-FETs.<sup>[15]</sup> This demonstrates the high performance of unfunctionalized graphene microtransistors which keep the ability to detect several ion species on the same device. To assess the dynamic of the G-FET response, the drain current was monitored while increasing KCl concentration in time (Figure 3b,c). The G-FET was polarized at  $V_{DS} = 80$  mV and the liquid reference was fixed at  $V_{LG} = 400$  or  $-500$  mV (resp. Figure 3b,c) to optimize the current response of G-FETs to variations of concentration (see Figure 3a). The initial solution was 200  $\mu$ L deionized water. Then successive amounts of KCl stock solution were added to increase the concentration of ionic charges in the solution (Figure 3b,c). At  $V_{LG} = 400$  mV (resp.  $-500$  mV), the majority of charge carriers are electrons (resp. holes). Each injection of a few  $\mu$ L of highly concentrated KCl solution induces a positive (resp. negative) change of the drain currents whose amplitude increases with the amount of injected KCl ions. The dynamic response of G-FET to KCl injection is characterized by two phases: one rapid increase of the drain current until a maximum transient value and a second phase of slow decrease to reach a stabilized current value. The maximum value corresponds to the G-FET response to an instantaneous injection of the stock solution, inducing a highly concentrated ion flux. The stabilized value is related to a permanent regime corresponding to a homogeneous ion concentration in the electrolyte. Both values are not proportional to the injected ion quantity or related ionic concentration. Indeed, it seems that variations of current increase as a logarithm of the ion concentration (Figure 3e, red circles). However, these current variations are comparable to what is expected by extracting drain current values from the transfer curves as a function of the KCl concentration, and at fixed gate potential  $V_{LG}$ , (Figure 3e, blue circles). This effect is directly related to the induced variations of potential at the graphene surface which is not linear as a function of the surface charge density but follows the Grahame equation:  $V_S = \frac{2k_B T}{e} \sinh^{-1} \left( \frac{en_s}{\sqrt{8\epsilon_0 \epsilon_r k_B T N_A 1000 C_{ion}}} \right)$ . From the polarity of the G-FET response which indicates a negative doping of the graphene sheet, we could assume the absorption or physisorption of the positively charged potassium ions on the graphene surface. This positive charge accumulation acts as a positive gate leading to an increase of the electron density and higher drain currents. These results further confirm the ability of G-FET to detect extracellular transient ion fluxes at the microscale and to provide dynamic monitoring of the electrolyte concentration.

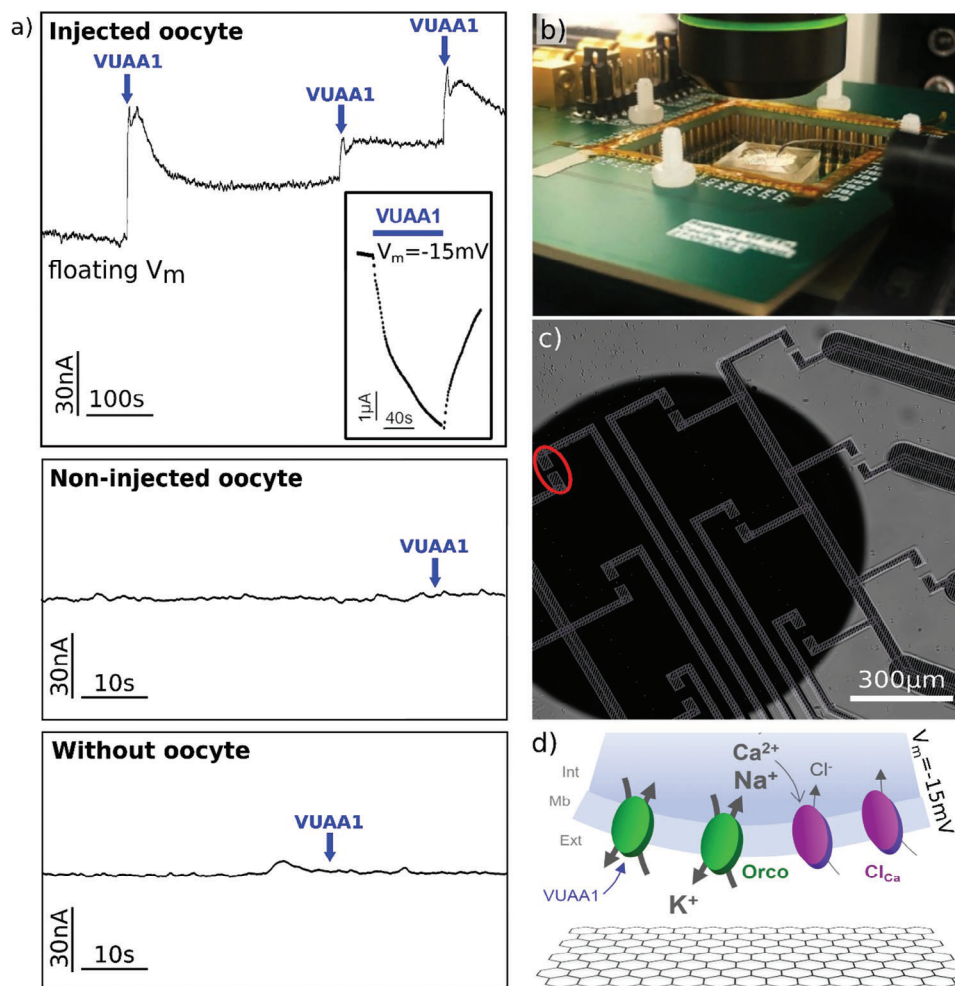
The choice of the reference electrode has an influence on the measured response to variations in the electrolyte ionic concentration. Indeed, pH and chloride concentration may have a direct influence on the value of the potential fixed by the Pt electrode. It can explain the small drift of the current baseline observed in Figure 3b. However, the setup composed of a G-FET and a gate electrode can be seen as a circuit of two equivalent capacitances corresponding to the EDLs at the graphene and Pt electrode interface. Considering that the dimension ratio of both electrodes is large ( $\frac{Pt \text{ electrode}}{\text{graphene channel}} \gg 1$ ), we can assume that the graphene is mainly sensitive to charge fluctuations occurring at the graphene surface. Moreover, ionic measurements have been performed in

a buffer solution to limit the electrode instability induced by pH variations. We have then assessed to which extent G-FETs can monitor biologically gated ion fluxes by coupling the G-FET array with *Xenopus* oocyte expressing ion channels and investigated the potential of such biological and electronic hybrid devices for chemical sensing.

### 2.3. Coupling G-FET with Non-Vertebrate and Artificial ICCRs

In non-vertebrate animals, insects possess a particular family of seven transmembrane ICs (7TMICs) that mainly compose their olfactory and gustatory systems.<sup>[33]</sup> Among these 7TMICs, the insect olfactory receptors (iORs) form the largest subfamily and are composed of two distinct subunits, a highly conserved Orco subunit that physically and functionally associates with odorant receptor (OR) subunits that evolved to recognize specific odorant molecules. Orco is able to form homotetrameric cation channels<sup>[34]</sup> and is activated by a few synthetic ligands such as the original one, VUAA1.<sup>[35]</sup> Orco generates high current amplitudes in response to ligand (VUAA1) in *Xenopus* oocytes,<sup>[5]</sup> and it was consequently used as a first model (Figures 4 and 5). On the other hand, artificial ICs can be engineered by fusing ICs with GPCRs.<sup>[6,7]</sup> GPCRs are not ICs but membrane receptors belonging to the largest family of human membrane receptors with more than 800 members and they are activated by a very large diversity of ligands from ions to proteins (e.g., hormones), as well as photons.<sup>[36]</sup> Thus the combination of ICCRs with G-FET has been later tested (Figure 6), as it could further provide suitable candidates for real-time detection and identification of various biomarkers or chemical compounds in general.

For each IC model (Orco and ICCR), heterologous expression of the ICs in defolliculated *Xenopus* oocytes was obtained by standard microinjection of mRNA and  $\geq 72$  h incubation at 19 °C in modified Barth's solution supplemented with penicillin, streptomycin, and gentamycin antibiotics. *Xenopus* oocytes have been placed above the FET arrays that are shown in Figures 4 and 6. The orientation of the oocyte over the FET is of particular importance due to the heterogeneous density of ion channels gradually concentrated around the injection point of mRNA.<sup>[37]</sup> To easily identify the injection point after incubation, all oocytes were micro-injected in the animal dark pole. The vitelline envelope surrounding the oocytes and made of glycoproteins was previously peeled off to reach the lipidic plasma membrane containing the ICs. This step was performed just before the experiment due to the high fragility of peeled oocytes, and contact with the air interface had to be strictly avoided to prevent the destruction of the membrane. The dark pole was carefully placed over the FETs devoted to recordings. The G-FET array and the oocytes were immersed in 300  $\mu$ L of a physiological buffer (ND96) containing high concentrations of sodium (91 mM), chloride (99 mM), and calcium (1.8 mM), and low concentrations of potassium (2 mM) preserving chemical gradients with intracellular concentrations ( $\approx 10$  mM Na<sup>+</sup>,  $\approx 44$  mM Cl<sup>-</sup>,  $\approx 10^{-5}$  mM Ca<sup>2+</sup> and  $\approx 109$  mM K<sup>+</sup>)<sup>[38]</sup> as described within Figure S3 (Supporting Information). A large polymethylsiloxane (PDMS) chamber assembled with the G-FET array allowed to contain the liquid and the quasi-reference Pt electrode used for applying the liquid gate voltage.



**Figure 4.** Real-time detection of Orco channel activation in *Xenopus* oocyte. (a) Timestamp of the drain current  $I_D$  during the three successive applications (30  $\mu\text{L}$ ) of Orco agonist VUAA1 (respectively 90, 183, and 260  $\mu\text{M}$ ). The same ligand application is repeated on non-injected oocytes (does not express the Orco protein) and G-FET without oocyte coupling. G-FET is polarized at  $V_{\text{DS}} = -20$  mV. The acquisition frequency is 25 kHz. Inset: Typical response of TEVC to VUAA1 activation of Orco channels. By convention, the current is negative, and an increase in the current amplitude results in a downward deflection. (b) G-FET electronic set-up under transmission microscopy. The reference potential of buffer electrolyte (ND96) is fixed with a quasi-Pt reference electrode at  $V_g = 0$  V. (c) The *Xenopus* oocyte is placed on a G-FET array using optical microscopy. G-FET used for Orco real-time measurements (a) is marked by a red circle. (d) Schematic showing the flow of ions through Orco (green) expressed in the plasma membrane of oocytes and when activated by VUAA1. Endogenous  $\text{Cl}_{\text{Ca}}$  channels are depicted in purple and activated by intracellular calcium.  $V_m$  is the transmembrane voltage of the *Xenopus* oocyte at the resting state.

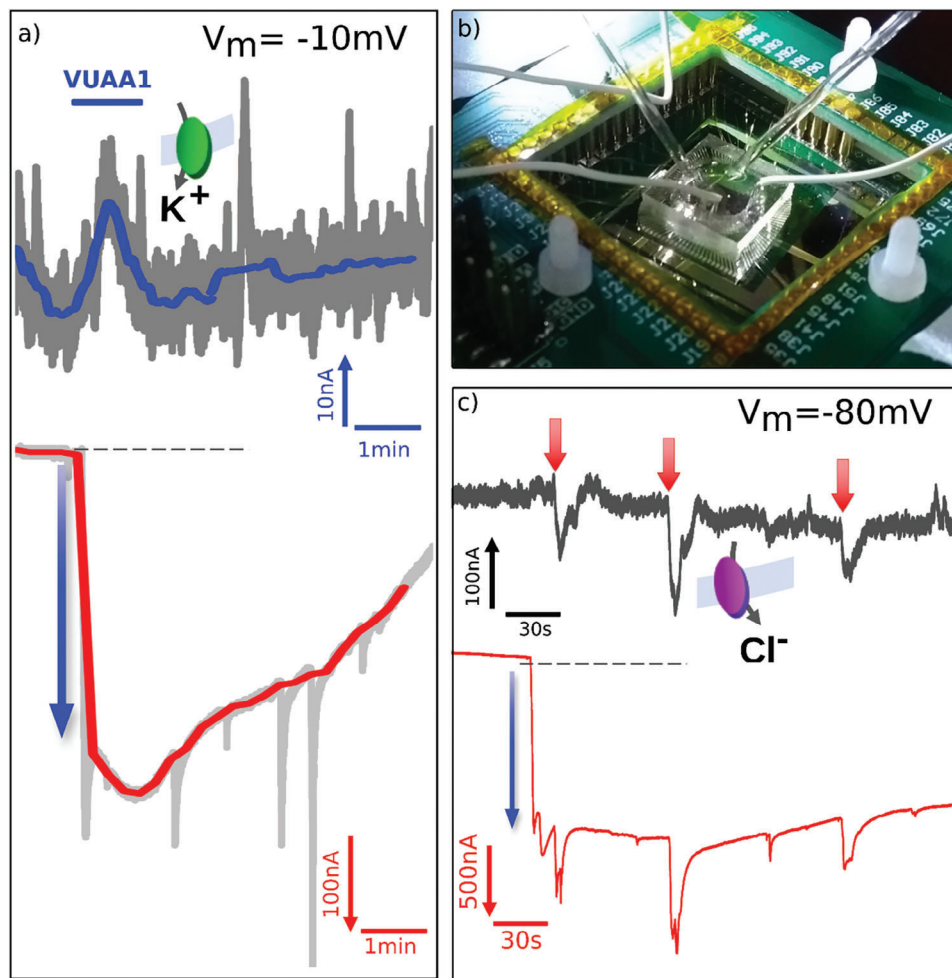
For all assays, the G-FET signal was recorded a few minutes after oocyte immersion to ensure the membrane adhesion to G-FET. The drain current was monitored in time while injecting small amounts of chemical compounds that are expected to activate or inhibit the opening of the engineered K channels. The recordings of Orco are shown in Figures 4–5 and Figure 6 for the engineered ICCRs.

#### 2.4. Real-Time Detection of Orco IC with G-FET

The time trace of the G-FET drain current has been monitored in real-time during the activation of Orco, and compared with two control conditions namely with oocyte but non-expressing Orco (non-injected oocyte), and without any oocyte (Figure 4).

The opening of Orco ICs occurs when adding VUAA1 ligand (typ. 0.1 mM) to the solution, which binds directly to the ICs. Figure 4a shows the drain current time trace after three injections of 30  $\mu\text{L}$  of 1 mM VUAA1 stock sequentially increasing the final concentration of VUAA1 from 90  $\mu\text{M}$  to 173  $\mu\text{M}$  to 250  $\mu\text{M}$ . These three successive injections result in clear increases of the drain current  $\Delta I \approx 23$ , 21, and 10 nA with a signal-to-noise ratio of  $S/N \approx 9$ , 8, and 4 respectively. For each detection of VUAA1, the response time of the GFET is  $\approx 2$  s, and no delayed response (more than 10 s) is expected. Figure S5 (Supporting Information) illustrates a zoomed view at the injection time to highlight the immediate response of the FET.

These significant increases are not visible on negative controls with oocytes that do not express Orco, nor on those without oocyte coupling confirming further the detection of Orco channel



**Figure 5.** Real-time measurements of G-FET and TEVC signals while activating Orco channels in *Xenopus* oocyte. (a) Representative coupled TEVC–G-FET recordings showing the currents induced by Orco agonist (VUAA1). The membrane potential is  $V_m = -10$  mV. G-FET is polarized at  $V_{DS} = 10$  mV. Both G-FET (blue) and TEVC (red) responses to VUAA1 application are characterized by an increase of measured current. Dash line corresponds to the TEVC current baseline (b) Coupled G-FET and TEVC measurement set-up. The membrane potential of the oocyte is monitored by TEVC. (c) Real-time recording of ionic currents with higher amplitude. TEVC current induced by the VUAA1 application is indicated by the blue arrow. Red arrows indicate transient ion currents.  $V_m = -80$  mV,  $V_{DS} = 20$  mV.

activation and the ability of G-FETs to measure transmembrane ion fluxes from few ion channels localized at the cell-device interface.

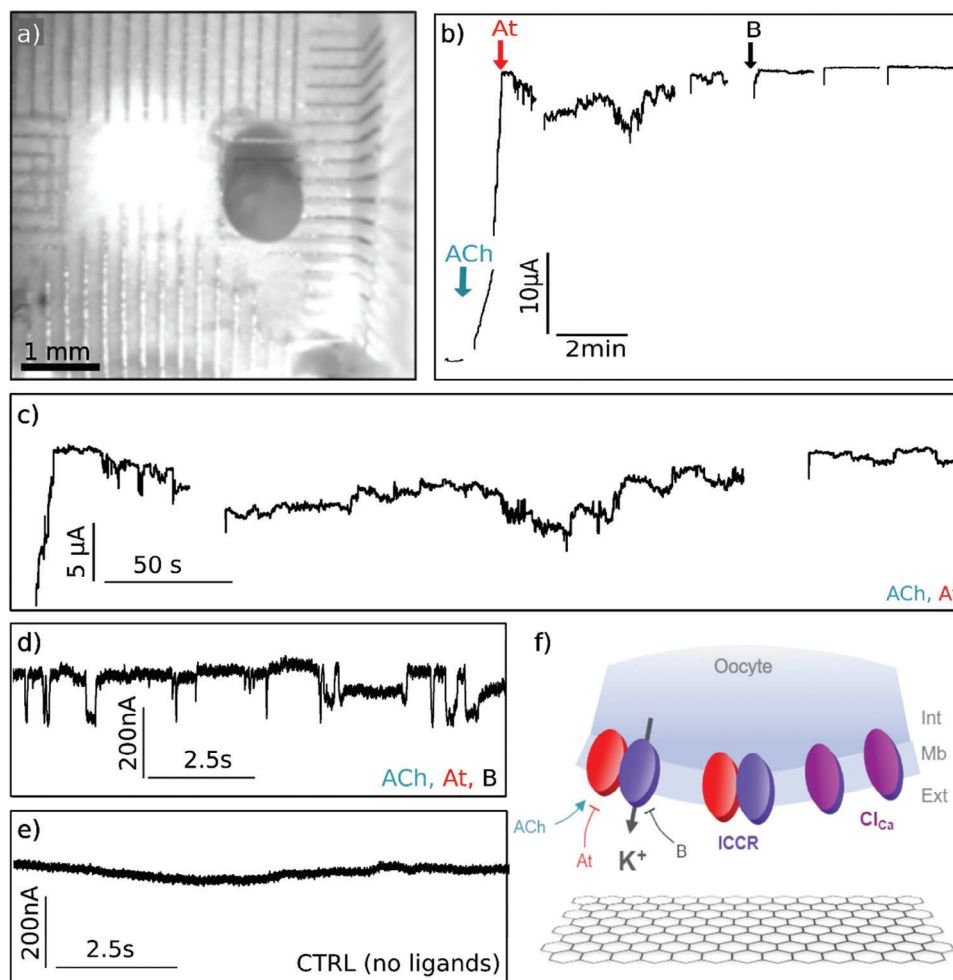
The opening of Orco ICs should induce an outward flow of  $K^+$  and an inward flow of  $Na^+$  and to a lesser extent  $Ca^{2+}$  (Figure 4d). The inward flow of micromolar concentration of  $Ca^{2+}$  could then activate endogenous calcium-activated chloride channels  $Cl_{Ca}^-$  (inward  $Cl^-$  fluxes) which is usually not observed in TEVC recording because the equilibrium potential of  $Cl^-$  ( $E_{Cl^-} = -20$  mV) is slightly higher or close to the measure  $V_m$ . Thus, the flow of  $Cl^-$  is usually negligible when biasing the membrane potential  $\approx -20$  mV (inset Figure 4a). However, such current could occur during G-FET measurement as the membrane potential is floating (Figure 4a) or when applying a higher negative value of the membrane potential (Figure 5c).

The graphene channel is slightly n-doped when interfacing with the oocyte (Figure S4, Supporting Information). In the electron regime, the polarity of the G-FET response ( $\Delta I_D > 0$ ,

Figure 4a) is as expected for sensing positively charged cations, which are presumably potassium here. Indeed, with the concentration of  $Na^+$  being significantly higher than  $K^+$  in the extracellular medium (91 mM NaCl and 2 mM KCl, see material and methods), an increase of  $K^+$  is expected to be easier to detect. Also, the G-FET signal is completely suppressed when blocking  $K^+$  current at  $V_m = -80$  mV (Figure 5c). Interestingly a double peak occurs for each injection which could stem from the activation of another type of ICs, possibly inward flow of  $Ca^{2+}$  that is present at low concentration in the extracellular bath (1.8 mM  $CaCl_2$ ).

Despite the double peak, the time response is similar to what was obtained by inducing artificial ion fluxes (see Figure 3a). The drain current increases first toward maximal values and then decreases toward a lower drain-current value after the complete diffusion and screening of the additional charges induced by the activations of the ICs. Note that the current step between VUAA1 applications slightly decreases while adding a new agonist.





**Figure 6.** Real-time detection of engineered K<sup>+</sup> channels expressed in Xenopus oocyte with G-FET. (a) Optical micrograph of the sensed oocyte which is directly positioned on a G-FET. (b) Successive timestamps of the drain current  $I_D$  during the successive application of acetylcholine (ACh), the endogenous ligand, then atropine (At, antagonist), and finally barium (B, K<sup>+</sup> channel blocker) as highlighted by the three arrows. (c, d) Zoom-views of the  $I_D$  time course, just after adding atropine (c) which is expected to inhibit the ACh activation of the ICCRs, and after the injection of barium (d) which suppresses all signals. (e) Control current time trace recorded prior to any injection on the same G-FET coupled with the oocyte. There is no variation in ion concentration at the G-FET interface. (f) Schematic showing the flow of potassium ions through the muscarinic M2 ICCR when activated by ACh. At is a potent antagonist that inhibits the activation of M2. B) is a generic blocker of potassium ions.

Indeed, a large number of Orco channels are open after the first agonist application. Following VUAA1 injections increase slightly the number of open ICs until it reaches a saturation state.

## 2.5. Dual TEVC and G-FET Recordings of Ligand-Gated Orco

To further assess the field-effect transduction of physiological ion fluxes, real-time TEVC and G-FET were performed simultaneously (Figure 5). The Xenopus oocytes were placed on a G-FET array immersed in a PDMS chamber containing 300  $\mu$ L of ND96 and were impaled by two pulled pipettes ( $\approx 0.22$  M $\Omega$ ) filled with 3 M KCl and an Ag/AgCl electrode measured against a reference electrode immersed in the ND96 buffer surrounding the oocytes (same condition than previously). Initially, the oocytes exhibited a resting membrane potential  $\approx -15$  mV and the membrane voltage was clamped at  $-10$  mV (Figure 5a) and  $-80$  mV (Figure 5c).

In Figure 5a, the simultaneous time traces of the TEVC and G-FET recordings reveal a small increase in drain current while adding 30  $\mu$ L at 1 mM of VUAA1 (90  $\mu$ M final at equilibrium state). The signal amplitude is comparable to previous measurements (Figure 4). However, the signal-to-noise ratio is lower, probably due to noise injection of TEVC coupling. The current measured by TEVC is also increasing toward negative values, which confirms the activation of Orco but indicates a dominant inward flow of positive charges across the cell membrane mainly induced by Na<sup>+</sup> ions, which are counterbalanced by an outward flow of K<sup>+</sup> ions (Figure S6, left panel, Supporting Information). Without the outward K<sup>+</sup> current, the TEVC signal is expected to be higher (few  $\mu$ A, inset Figure 4a). This result indicates that G-FET dominantly detects the increase of K<sup>+</sup> ions in the extracellular space (which contains only 2 mM of K<sup>+</sup>,  $\Delta C_K = +107$  mM) that is proportionally higher than the depletion of Na<sup>+</sup> from the buffer (which already contains 96 mM of Na<sup>+</sup>). It should be noted that the

current variations are much more significant (20 times higher) in the TEVC than in the G-FET recording. While the current is collected from the whole cell during TEVC recording, the reduction of the sensing area when using G-FET (from  $\approx 1 \text{ mm}^2$  to  $200 \text{ }\mu\text{m}^2$ , respectively) should reduce the number of detectable ICs and the amount of potassium ions detected by G-FET. The variation of drain-current is thus highly dependent on the IC heterologous expression rate. Considering a typical density of  $10^5 \text{ IC}\cdot\text{mm}^{-2}$ , the number of ICs in the vicinity of the graphene surface should be  $\approx 10$  ICs (vs  $\approx 10^5$  for whole-cell TEVC recordings). A slight variation in the ion channel density could significantly impact the detection efficiency of the G-FET measurement. Although TEVC exhibits a clear response to ligand activation, the G-FET detection of ion currents is not reachable on every measured oocyte, depending mainly on the expression rate of ICs, the density of ICs coupled with the graphene channel, and the quality of the interface after successive uses.

When clamping the voltage at  $-80 \text{ mV}$  (Figure 5c), the outward  $\text{K}^+$  currents are blocked. As a result, the amplitude of the TEVC current is increased ( $330 \text{ nA}$  at  $-10 \text{ mV}$  to  $1500 \text{ nA}$  at  $-80 \text{ mV}$ ), which is expected with a higher inward flow of  $\text{Na}^+$  and  $\text{Ca}^{2+}$  and a lower outward flow of  $\text{K}^+$  (red trace Figure 5c, blue arrow). The absence of response when adding the ligand (black curve Figure 5c) confirms that the G-FET is rather sensitive to the outward potassium current than to the sodium entry. Interestingly, after a few minutes, the G-FET response reveals drain current variations with opposite polarity (Figure 5c, red arrows). This inverse polarity agrees with a rapid increase of  $\text{Cl}^-$  ions in the extracellular space when the membrane voltage is clamped at  $-80 \text{ mV}$  (Figure S6, right panel, Supporting Information). *Xenopus* oocytes endogenously expressed  $\text{Ca}^{2+}$ -activated chloride ( $\text{Cl}_{\text{Ca}}$ ) channels that generate large  $\text{Cl}^-$  currents when the concentration of intracellular calcium ions increases in the micromolar range ( $\text{EC}_{50} \approx 25 \text{ }\mu\text{M}$ ).<sup>[39]</sup> When Orco channels are activated, a large flow of  $\text{Ca}^{2+}$  enters the cell due to the high gradient and the high negative membrane voltage, and  $\text{Ca}^{2+}$  ions activate these  $\text{Cl}_{\text{Ca}}$  channels. Due to the important density of these channels especially in the dark animal pole in contact with graphene,<sup>[39]</sup> and their sensitivity to intracellular calcium and the weak outward flow of  $\text{K}^+$  at  $V_m = -80 \text{ mV}$ ,  $\text{Cl}^-$  accumulation in graphene vicinity is dominant.

The nature of the signals is also consistent with the inversion of polarity on G-FET measurements. A  $\text{Cl}^-$  accumulation at the graphene-cell interface induces a positive p-doping of the graphene channel, reducing the G-FET channel conductance in the electron regime. Here, G-FET exhibits a clear response to  $\text{Cl}^-$  flow. The recordings enable us to identify the ion polarity of transmembrane currents (positive out or negative in) thanks to signal polarity, which is not the case for TEVC. Indeed, negative variations of TEVC current are induced either by positive charges flowing inward the cell or negative charges going outward.

Conformational rearrangements of the ICs could barely contribute to the change in G-FET response. In the current configuration, the G-FET detects only the changes occurring on the extracellular side of the proteins. Upon ligand binding, receptors do change their conformation, however, the conformational changes are mostly intracellularly for the GPCRs and only by rotation of helices for Orco. The density of charge is unaffected by these

changes and cannot explain the amplitude of signals recorded by TEVC and G-FET that could only rely on large flow of ions.

## 2.6. Application for the G-FET Detection of ICCRs

The designed ion channel coupled receptor is activated by muscarinic M2 receptor agonists such as acetylcholine (ACh) and antagonized by compounds such as atropine (At). The signal is generated by the opening of the coupled potassium channel and it is blocked by barium (B). The experiments were performed on the G-FET setup (Figure 6a) that was not parallelized with the TEVC setup. The drain current was monitored in time while injecting  $3 \text{ }\mu\text{L}$  of ACh, At, and B, which are expected to activate and inhibit the opening of the coupled  $\text{K}^+$  channels, and then block all potassium channels respectively. Figure 6 shows successive timestamps of  $\approx 1$  to  $5 \text{ min}$ , which correspond to acquisitions performed at a sampling rate of  $4 \text{ kHz}$ .

Prior to any injection, the baseline value of the drain current is  $\approx 53 \text{ }\mu\text{A}$  ( $t = 50 \text{ s}$  Figure 6a, with  $V_{\text{DS}} = 90 \text{ mV}$ , at  $V_{\text{LG}} = 0 \text{ V}$ ) which is as expected from the  $I_{\text{D}}-V_{\text{LG}}$  curve performed just before placing the oocyte (Figure S4, Supporting Information). The addition of ACh ( $5 \text{ }\mu\text{M}$  final) induces a sharp increase of the drain current  $\Delta I/I_0 \approx 0.6$  and a constant rate of  $\approx 0.8 \text{ }\mu\text{A s}^{-1}$  (Figure 6b). This natural ligand binds and activates the human M2 muscarinic receptor, which opens the coupled potassium channel (Kir6.2) and generates transmembrane potassium outward flow. This resulting  $\text{K}^+$  flux toward the extracellular medium modulates the underlying G-FET conductance. The polarity of the G-FET response ( $\Delta I_{\text{D}} > 0$ ) is the same as previous detection with Orco ICs. This observation validates both the expression of the receptor-coupled ion channel within the oocytes and the efficient detection of the ICCRs with G-FET.

The subsequent addition of atropine—a potent M2 receptor antagonist—immediately blocked almost totally the activation of the receptor that returned to a state close to its basal state and induced the same change of state of the coupled- $\text{K}^+$  channel. The drain current remained relatively constant few seconds before fluctuations in lower values. These variations, ranging from  $500 \text{ nA}$  to  $1-5 \text{ }\mu\text{A}$ , could result from the activation of endogenous  $\text{Na}^+/\text{K}^+$  ATPases that extracellularly export  $\text{Na}^+$  and intracellularly import  $\text{K}^+$  ions and certainly participate in the variations of the density of positive charges in the inter-membrane/G-FET space in coordination with IC activity. The last injection of barium ( $3 \text{ mM}$  final) blocked the potassium channels and suppressed all fluctuations. The baseline of the drain current remained constant up to the end of the experiment ( $I_{\text{D}} = 89 \text{ }\mu\text{A}$ ). Note that the time trace still exhibited step-like variations  $\Delta I_{\text{D}}$  ranging between  $70$  and  $130 \text{ nA}$  (Figure 6d). These fluctuations are not observed prior the activation of the K-channels by ACh (Figure 6e) and could stem from the flickering noise of a few ICCRs since the rectangular-type signals are similar in single ion channel recordings.

From the amplitude of the drain current, we could estimate the amount of ICs sensed by the G-FET. The following equation enables to extraction of the equivalent variation of gate voltage within the cleft between the oocyte and the graphene channel  $\Delta V_{\text{g}} = \Delta I_{\text{D}} \times (g_{\text{m}} V_{\text{D}})^{-1}$ . Considering the normalized transconductance  $g_{\text{m}}$  ( $1 \text{ mS/V}$ ) and the drain-source voltage  $V_{\text{DS}}$  ( $90 \text{ mV}$ ),

the equivalent variation of the gating potential  $\Delta V_g$  in the cleft ranges between 5 and 15 mV (for drain current steps of 0.5 and 1.5  $\mu\text{A}$ ) after injecting the atropine. For low surface potential, the Grahame equation provides the equivalent number of surface charges (or ions) by  $\Delta n = (A\epsilon_c\epsilon_0/\lambda_D) \times \Delta V_g$ . This value is ranging here between  $\Delta n = 0.6$  and 1.8 pC, with the exposed graphene area and Debye length being  $A = 200 \mu\text{m}^2$  and  $\lambda_D = 1 \text{ nm}$ . Single ICCR current being  $\approx 1\text{--}3 \text{ pC s}^{-1}$ , the amplitudes of the step-like variation are indeed consistent with the opening/closing of individual ICCRs.

The results with ICCRs further demonstrated the high sensitivity and versatility of the G-FET detection for sensing transmembrane current in biological systems, which is a key building block for implementing advanced bio-hybrid electronics.

### 3. Discussion

Here, the reported sensitivity and limit of detection have been shown to be comparable to state-of-the-art functionalized graphene devices, but with an additional advantage which is the ability to measure several ion species (e.g.,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Cl}^-$ ) in real-time with a same G-FET. This feature is crucial for electrophysiological recording and for advanced biohybrid electronics. Indeed, this first demonstration of combining graphene ISFET with biological ICCRs opens an avenue of investigation, regarding the available number of human ICs that might be provided by the ICCRs (up to 800 members).

Graphene bioelectronics has demonstrated efficient sensing properties enabling to detection of few to single ligand-activated ICs distributed on very localized surfaces of the oocyte membranes. However, a current drawback of the field-effect detection with graphene remains the significant number of undetected events (ligand-activation of ICs) which could be an issue for high throughput biosensing. To overcome this possible limitation, several features might be optimized to provide the required detection rate and reproducibility. Here, we discuss several device parameters that could be improved in future development, such as the detection threshold, the sensitivity, the signal-to-noise ratio, and the quality of the interface.

Because graphene surface charges are rapidly screened by charges in bulk solution, the limit of detection of the reported G-FETs is quite high when sensing ionic species in large volumes ( $\approx 7.10^{-4} \text{ M}$  and equivalently  $1.10^{-7} \text{ mol}$  in 200  $\mu\text{L}$ , Figure 3). However, strong coupling between the G-FET and the ion-channel receptor, such as in *Xenopus* oocytes, enables us to sense weak variations of ion concentration in time (being  $\approx 10^{-16} \text{ mol s}^{-1}$  for 20 opened ICs). Detecting local changes of ion concentration in a solution containing a high concentration of ions would be not possible if the cell membrane does not adhere to the substrate. Adhesion of the cell induces a seal which creates a significant resistance with the grounded buffer solution (Figure S1, Supporting Information). The opening of the ion channel generates ion currents within an ultrathin liquid junction between the cell membrane and the G-FET (typically  $<10 \text{ nm}$ )<sup>[40]</sup> that promote the G-FET detection efficiency.

Thus, the sealing resistance at the cleft between the oocyte and the graphene channel is expected to improve the detection sensi-

tivity. In this closed configuration, small variations of ion amount are easier to detect than in an opened media. Although this apparent high detection sensitivity, the response of the G-FETs could be further improved by tuning the G-FET operation regime toward more sensitive setpoints (e.g.,  $V_{\text{LG}} = -500 \text{ mV}$ , Figure 3a) without altering the physiological properties of the oocytes and the ICs to-be-sensed.

Beyond the amplitude of the G-FET response, the electrical noise is an important feature to improve the number of detected events. For G-FET recordings, the typical noise amplitude is  $\approx 3 \text{ nA}$ , enabling a good signal-to-noise ratio ( $S/N \approx 9$ ) while detecting ionic currents (Figure 4). However, the noise amplitude was significantly increased when coupling G-FET to TEVC (Figure 5), reducing further the detection efficiency. A dedicated setup that will reduce the background noise during dual G-FET and TEVC recording, is thus expected to further enhance the detection rate.

On the other hand, the G-FET detection efficiency is expected to be lower than whole-cell TEVC measurements as it mainly depends on the probability of coupling G-FET with a well-activated IC. The ligand application might trigger the opening of ICs which are located away from the transistor channel and not detected by the G-FET. To overcome this limitation, experimental conditions can be improved on both biological and electrical aspects. To that end, the expression rate of ICs has been optimized, being  $\approx 1.10^5 \text{ IC.mm}^{-2}$ . Moreover, the oocytes are interfaced with G-FET by coupling the pole with higher expression probability, the dark pole of the oocyte. Despite this precaution, the number of coupled ICs is expected to be significantly reduced, being  $\approx 10$  for a  $10 \times 20 \mu\text{m}^2$  device if all ICs are activated and if the ICs density is maximal and homogeneous above the G-FET. For such a low quantity, the stochasticity of the ICs expression or activation could no longer be negligible. Thus, the number of activated ICs above the G-FETs is probably less than 10 ICs. While reducing the device size might be a strategy for sensing individual or few ion channels, some applications may require higher efficiency. In that case, the ICs are also more likely to be sensed by increasing the effective surface of the graphene-membrane coupling. Larger graphene devices can be designed to increase the detection surface and the probability of interfacing ligand-activated ICs with the graphene channel. Then, the G-FET dimensions should be carefully chosen to keep the required sensitivity which is proportional to the W/L ratio, W and L being respectively the width and length of the transistor.

Here, five independent G-FETs were used to provide the first demonstrations within liquids and cells. Future investigation should be done to optimize the demonstration in the showcase application, providing a statistical study of the sensing performance and assessing the robustness of the calibration which are both crucial for chemical detection.

Another asset of arrays of micrometer-size G-FET is multisite and multi-cell recording. Hence several areas of the oocytes or several oocytes could be measured at the same time, which optimizes the detection probability. Additionally, G-FET arrays can be coupled to microfluidic circuits to control the oocyte/G-FETs coupling and liquid injections for drug applications or medium changes.

Lastly, the chemistry at the graphene-liquid interface is suspected to play a key role in the detection efficiency and

repeatability. The hydrophobic monolayer graphene strongly interacts with the surrounding proteins, as well as with the cell membrane. The absorption of phospholipids at the graphene surface could act as a physical barrier for the ions to be detected. Also, the presence of residual chemicals such as IC inhibitors (e.g., At, barium) at the graphene surface will block consecutive activations of ICs, thus no signal will be detected when repeating the experiment with the same G-FET. Thus, particular care will be taken in future works to assess the extent of such interaction on the detection rate. A dedicated rinsing protocol and functionalization strategy could then be implemented to optimize the interaction with the ions to be detected while minimizing the reaction with the other surrounding biological or chemical species.

## 4. Conclusion

The reported works demonstrated the high sensitivity of the G-FET for the detection of ultralow variation of ion concentration in saline media. The reachable detection threshold being about few to single potassium  $K^+$  ICs, G-FETs outperform conventional microdevices while keeping optical access to the recording site for interfacing soft and living matter. To that aim, the graphene field effect transistors have been successfully assembled with engineered *Xenopus* oocytes for the detection of ligand-activated ICs. The G-FETs have been shown to efficiently sense the evoked activity of ICs by ligand application in-operando (cell culture solution) and for two classes of ICs (Orco and ICCR). Dual acquisitions with TEVC and G-FET demonstrated that a G-FET enables efficient transduction of ion fluxes into readable electronic signals, as clear similitudes have been observed in the time traces of the TEVC and G-FET signals. The electronic properties of liquid-gated G-FETs allowed quantitative measurements in a high-frequency regime that resolved a broad dynamic of ICs opening and closing. Beyond the detection of transmembrane current ions, G-FET measurement could allow to identification of the polarity of the sensed ions, providing additional cues on the nature of the ion fluxes to complement TEVC measurements. Finally, the micrometer size of G-FET enabled local measurement of a few individual ICs, in addition to being sensitive to whole cell current variation. The multisite recording ability offered by the G-FET array also has numerous possibilities for multiscale sensing, which is of both fundamental and applied interests such as providing a functional map over a single or assembly of cells, or for high throughput screening of cellular solutions or analytes. Although the promising sensing properties of G-FET have been reported, the measurement repeatability should be further improved to bring this novel kind of biohybrid device as the next golden standard for ICs recordings.

## 5. Experimental Section

**G-FET Fabrication and Characterization:** Monolayer graphene was obtained by CVD and transferred from the copper growth substrate onto sapphire by wet transfer technique. The front side of the copper foil supporting the monolayer was protected by a thin layer of poly(methyl methacrylate) (PMMA) resist, while the backside was successively etched in oxygen plasma (few seconds) and ammonium persulfate (1 mM, for  $\approx 1$  h). After complete copper etching, the floating graphene/PMMA bilayers were

rinsed in several baths of deionized water to remove all solvent traces and then gently fished with the final substrate (to be used for the sensing experiments). The sample was left 1 h under a laminar flow hood to dry and finally heated up to 180 °C to remove water and improve the adhesion of the monolayer graphene on the substrate. The graphene channels were then obtained by patterning a positive resist with laser lithography and dry etching in  $O_2$  plasma. The metallic drain and source contacts were obtained by wet etching of metals (1 nm Ti, 30 nm Pd, 20 nm Au) designed using a negative resist mask. These metallic lines were finally passivated with S1818 resist to insulate them from the liquid gate (Figure 2a). Figure 2 shows a typical sample and zooms on a G-FET to illustrate the high quality of the resulting graphene channels directly exposed to the analytes. Only a few residuals of the fabrication process are observed, and wrinkles can already be observed as highlighted by the arrows.

**Expression of the ICs in *Xenopus* Oocytes:** Orco gene from *Drosophila melanogaster* was codon-optimized, synthesized, and subcloned in a derived pGEMHE vector by Genscript. pGEMHE was optimized for protein expression in *Xenopus* oocytes as previously described.<sup>[6,30]</sup>

To design ICCR, the genes of human M2 and mouse Kir6.2 were fused by PCR by insertion of the GPCR coding sequence at the 5' end of the Kir6.2 gene cloned in the derived-pGEMHE vector.

DNAs coding for Orco and ICCR were amplified using Qiagen MidiPrep Kit, linearized in the 3' regions of the polyA tail, purified by standard phenol:chloroform extraction and transcribed in mRNA using the T7 mMessage mMachine Kit (Thermo Fisher Scientific) following the supplier's protocol. mRNAs were purified by the standard phenol:chloroform protocol, analyzed by agarose-gel electrophoresis, and quantified by spectrophotometry.<sup>[33]</sup> For the heterologous expression,<sup>[26]</sup> *Xenopus* oocytes were defolliculated by collagenase treatment and microinjected with 50 nL of RNAs-e-free water containing 20 ng of Orco or 5 ng of M2-Kir6.2 cRNAs. Oocytes were incubated in individual wells in Barth's solution supplemented with penicillin-streptomycin and gentamicin mix for at least 48 hrs at 19 °C. The recording buffer (ND96) was composed of 91 mM NaCl, 2 mM KCl, 1.8 mM  $CaCl_2$ , 1 mM  $MgCl_2$ , 5 mM HEPES, and pH 7.4. In ICCR recording, 0.3 mM niflumic acid was added to the buffer to partially block endogenous  $Cl^-$  currents. Before loading oocytes on G-FET, the vitelline envelope of *Xenopus* oocytes was manually removed with sharp forceps (Moria 9980) under a stereomicroscope (Leica MZ6) and in ND96 buffer.

All chemicals were purchased from Sigma-Aldrich. Stock solutions were acetylcholine chloride 5 mM (water), atropine 1 mM (ethanol),  $BaCl_2$  1.5 M (water), and VUAA1 108 mM (DMSO).

**TEVC and G-FET Sensing Recordings:** For parallelization of G-FET and TEVC recordings, the G-FET array and the electronic card were placed in a standard manual TEVC setup in a Faraday cage and under a stereomicroscope (Leica Wild M3Z). In a PDMS chamber containing 300  $\mu$ L of ND96 buffer, the peeled oocytes were placed with a Pasteur pipette on a connected G-FET that was electrically characterized before oocyte addition. Using micromanipulators, the oocytes were impaled by two pulled borosilicate pipettes (WPI TW150F-4,  $\approx 0.22$  M $\Omega$ ) filled with 3 M KCl and an Ag/AgCl electrode. Reference electrodes were immersed in the ND96 buffer surrounding the oocytes. The TEVC setup was composed of the Geneclamp 500B amplifier (Axon), and the Digidata 1440A (Axon) and controlled with Clampex 10.44 software (Axon). Recordings were performed at 10 kHz and filtered at 10 Hz with a voltage clamped at  $-10$  and  $-80$  mV (Figure 5a,b respectively). Ligands were applied successively at the appropriate volumes with a standard micropipette. The electronic card allows to record simultaneously two independent G-FETs on the same chip. Ion-sensing within the KCl solution was performed on two independent G-FETs. Three independent G-FETs were used for ion-sensing within living cells.

## Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

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## Conflict of Interest

The authors declare no conflict of interest.

## Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

## Keywords

bioelectronic, biosensor, chemical detection, field-effect-transistors, graphene, ion channel, Xenopus oocyte

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