

Stability of the tetragonal phase of BaZrO₃ under high pressure

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In this paper, we revisit the high pressure behavior of BaZrO₃ by a combination of first-principle calculations, Raman spectroscopy and x-ray diffraction under high pressure. We confirm experimentally the cubic-to-tetragonal transition at 10 GPa and find no evidence for any other phase transition up to 45 GPa, the highest pressures investigated, at variance with past reports. We reinvestigate phase stability with density functional theory considering not only the known tetragonal (*I4/mcm*) phase but also other potential antiferrodistortive candidates. This shows that the tetragonal phase becomes progressively more stable upon increasing pressure as compared to phases with more complex tilt systems. The possibility for a second transition to another tilted phase at higher pressures, and in particular to the very common orthorhombic *Pnma* structure, is therefore ruled out.

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I. INTRODUCTION

The family of perovskite oxides is known for its large variety of structural distortions that is, in turn, crucial for their physical properties. The most common distortions are collective rotations (or tilts) of the corner-sharing oxygen octahedra that lower the crystal symmetry from a simple cubic structure in the absence of tilt to a lower symmetry determined by its tilt pattern. Such oxygen octahedra rotations are typically referred to as antiferrodistortive (AFD) distortions. There is a long history of studies of those tilts [1], their classification [2–4], and how they evolve with external parameters. In the most simple tilt systems, octahedra rotate only around one of the pseudocubic axes and the rotations in two adjacent layers perpendicular to the rotation axis can be either in-phase ($a^0a^0c^+$, in Glazer's notation [2]), producing a phononic instability at the *M* point of the cubic Brillouin zone (BZ), or antiphase ($a^0a^0c^-$), with the instability appearing at *R* point [5]. General rules for the evolution of tilts under hydrostatic pressure have been formulated [6–10]: For most $A^{2+}B^{4+}O_3$ perovskites, it is

established that the tilt angle should increase and, if starting from a cubic structure, cause a phase transition towards a low-symmetry tilted phase. This has been investigated very early on with studies of the cubic-to-tetragonal ($Pm\bar{3}m \rightarrow I4/mcm$) phase transition in the classical perovskite SrTiO₃ [6] and confirmed in many subsequent studies (Refs. [11–13] and references therein). A similar scenario has been observed in other cubic perovskites including the oxide BaZrO₃ [14,15] and the fluorite KMnF₃ [16,17].

As much as this first phase transition under pressure is known, the behavior at higher pressures is still very unclear. It is natural to expect that the tetragonal phase observed under pressure in SrTiO₃, KMnF₃, or BaZrO₃ becomes unstable above some critical pressure, i.e., when the tilt angle reaches some critical value. This intuition partly originates from the observation of phase sequences with varying temperature that may display multiple transitions bridging different tilt systems. Particularly relevant examples are CaTiO₃ [18] and SrZrO₃ [19], which undergo phase sequences with increasing temperature $Pnma \rightarrow I4/mcm \rightarrow Pm\bar{3}m$ and $Pnma \rightarrow Imma \rightarrow I4/mcm \rightarrow Pm\bar{3}m$ respectively. In both cases, the cubic-to-tetragonal transition is continuous and involves a single tilt angle whereas the tetragonal-to-orthorhombic transition is discontinuous. Even richer sequences may be found in perovskites with competing polar instabilities such as NaNbO₃ [20,21]. Since pressure induces much stronger reduction in bond length than temperature, we might expect

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cubic \rightarrow *tetragonal* \rightarrow *orthorhombic* phase sequences for BaZrO₃, SrTiO₃, and similar perovskites under pressure.

BaZrO₃ is a particularly relevant model system to study the evolution of tilts under pressure. It is experimentally cubic at ambient pressure and down to 2 K, but hosts a tiny instability associated to AFD distortions revealed by density functional theory (DFT) that has been attracting a great deal of attention from a theoretical point of view [22–26]. This stirred discussions about its true ground state and the possible consequences of this instability on physical properties, such as the existence of locally distorted nanodomains, which were recently observed by electron diffraction and pair distribution function [27] or the anomaly of its dielectric constant [22,23]. Besides, unlike in SrTiO₃ and CaTiO₃ [28,29] where the polar instability at the zone center plays a major role, only the zone boundary tilt mode is unstable in BaZrO₃, which makes it a comparatively “pure” tilt system. Finally it has been shown that several AFD phases are nearly degenerate at ambient pressure [30], specifically the tetragonal $I4/mcm$ ($a^0a^0c^-$), the orthorhombic $Imma$ ($a^-b^0a^-$) and the rhombohedral $R\bar{3}c$ ($a^-a^-a^-$). This situation leaves the energy landscape particularly open for the stabilization of multiple phases and the existence of multiple phase transitions under pressure.

Experimentally, it was indeed proposed in several instances that a second phase transition occurs both in SrTiO₃ and BaZrO₃ [11,13,14,31], but the experimental results remain controversial. In SrTiO₃, the existence of a second phase transition has been hypothesized based on the observation of a Raman peak splitting [11], or changes in the Ti pre-edge features in x-ray absorption measurements [32,33]. However, this proposition was discarded by Raman spectroscopy and single crystal x-ray diffraction measurements [12]. In BaZrO₃, splitting of Raman peaks was also observed and invoked to propose a transition to an orthorhombic phase, also inspired from phases found in the phase diagram of the (Ba,Ce)ZrO₃ system [14], and a recent Raman study concluded for a $Pm\bar{3}m \rightarrow R\bar{3}c \rightarrow I4/mcm$ phase sequence [31]. Nevertheless, this was not confirmed by the high-pressure x-ray study in Ref. [15], where authors report evidence of a *cubic* \rightarrow *tetragonal* phase transition at about 17 GPa and persistence of the tetragonal phase up to ~ 46 GPa at room temperature. At the theoretical level, we are not aware of studies of the energy competition between different AFD phases in BaZrO₃; only predictions of pressure-induced softening of polar and antiferrodistortive modes have been proposed in Ref. [34], as also reported for other perovskite systems [35,36]. Also, the possibility for transitions to completely different polymorphs has been proposed [37,38]; in particular, a transition to a postperovskite phase was predicted to occur at pressures as low as 70 GPa and cause a first-order transition into a semiconducting orthorhombic phase with a 7.9% volume collapse. These predictions have not yet found experimental confirmation.

In this paper we revisit the behavior of BaZrO₃ under hydrostatic pressure with the aim to clarify its pressure-induced phase sequence. We show experimental data by Raman spectroscopy and x-ray diffraction (XRD), and discuss their interpretations also relying on supporting first-principle calculations of the dynamical properties and energetics as a function of pressure. The similar case of SrTiO₃ is also

discussed for comparison. Our study provides strong arguments for excluding other structural transitions than the cubic-to-tetragonal up to 45 GPa, and provides explanations for the divergent claims found in the literature.

II. METHODS AND EXPERIMENTAL DETAILS

Single crystal samples were prepared from a crystal purchased from Crystal Base Co. Ltd. and grown by the tri-arc Czochralski method, as described and characterized in Ref. [39].

We performed single crystal Raman measurements under pressure up to 20 GPa in a membrane-type diamond anvil-cell (DAC) with a 600 μm culet size using a 4:1 methanol-ethanol solution as pressure-transmitting medium. Pressure was measured in situ with two ruby balls inserted in the chamber together with the BaZrO₃ single crystal sample. The two ruby balls indicated the same pressure up to the highest pressures showing good hydrostaticity. The spectra were recorded using a 532 nm laser line and a Renishaw inVia confocal Raman spectrometer calibrated using the 521 cm^{-1} phonon mode of silicon.

Two sets of powder x-ray measurements were performed at the ID27 beamline of the ESRF synchrotron facility. Both measurements were performed using neon as a pressure transmitting medium. A monochromatized x-ray radiation source of wavelength 0.3738 Å and a two-dimensional charge-coupled device detector (MAR-CCD) with a pixel size of 79 μm were used. A silicon powder standard was used to calibrate the distance of the detector and other geometry parameters. For the first set, powdered BaZrO₃, referred to here as sample S1, was loaded on a DAC with a culet size of 300 μm and diffraction patterns up to 16.5 GPa with pressure steps of 0.7 GPa were obtained. The second set of measurements were performed up to 45 GPa using a DAC with a culet size of 250 μm and a second powder sample of BaZrO₃, called S2, prepared as before. The pressure was determined from the ruby calibration method [40] and the standard deviation in pressure measurement was of less than 1%, ensuring a good hydrostaticity [41]. The data was analyzed by full Rietveld refinements using FULLPROF software [42]. The refined parameters are: (i) the scale factor, (ii) manually chosen linearly interpolated background at 15 points, (iii) lattice parameters, (iv) fractional coordinates, (v) isotropic thermal parameters, and (vi) shape and broadening parameters. The peak shape was described with a pseudo-Voigt function. The profile parameters u , v , w , lx , and ly , which determine the resolution function (see FULLPROF manual), were obtained from the refinement of a high-purity silicon standard.

For single crystal XRD measurements, two small crystals with different orientations were loaded into a DAC and gave essentially identical results; the corresponding data will be labeled as sample S3 in the following. Measurements were performed at the CRISTAL beamline at the SOLEIL synchrotron facility. A monochromatized x-ray radiation source of wavelength of 0.41579 Å and a two-dimensional charge-coupled device (2D CCD, Rayonix SX165) with a pixel size of 79 μm and a diameter of 165 mm were used. A ruby single crystal and lanthanum hexaboride (LaB₆) powder were used to calibrate the geometrical parameters of the diffractome-

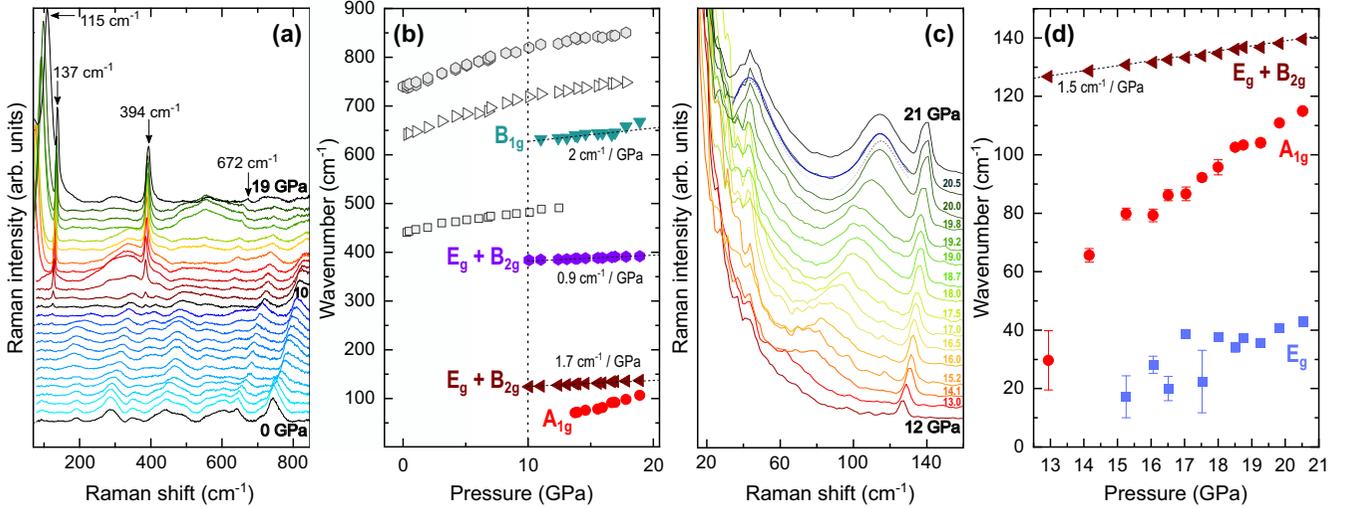


FIG. 1. Raman spectra of BaZrO₃ single crystals under pressure (a) from 0 to 19 GPa when increasing pressure, and (c) between 21 and 12 GPa when releasing pressure. The spectra in (a) are classical Stokes spectra, while the spectra in (c) are collected without cutting off the Rayleigh scattering of the laser line, allowing for observation of lower energies. An example of the Lorentzian fits performed on the Raman modes is shown in blue. (b) Pressure dependence of the wavenumbers of the phonon modes derived from the spectra obtained from 0 to 19 GPa, the solid symbols show the modes appearing above the transition at 10 GPa. (d) Behavior of the two soft modes (and first hard mode) appearing above 10 GPa, these frequencies are derived from the spectra shown in (c), fitted with Lorentzian functions.

ter and the wavelength respectively. The pressure inside the 600 μm culet-sized DAC was determined from the diffraction of powdered gold included near the samples and its reported equation of state (EoS) [43,44]. Helium was used as pressure-transmitting medium. The experiment was conducted up to 12.5 GPa at which point the diamonds got damaged, impeding to carry out a full structural refinement at higher pressures.

First-principle calculations of structural and dynamical properties rely on density functional theory (DFT) and density functional perturbation theory (DFPT) [45,46], as implemented in the ABINIT package [47–49]. The exchange-correlation potential was evaluated within the generalized gradient approximation (GGA) using the Wu-Cohen (WC) functional [50], which provides us with good description of the structural properties of the cubic phase and a lattice parameter a_0 of about 4.184 \AA — *cf.* experimental $a_0 \simeq 4.195 \text{\AA}$ [Fig. 2(c)] and lattice dynamics reported in our previous paper [30]—correcting the reported overestimation within the more conventional Perdew-Burke-Ernzerhof (PBE) functional [25] or underestimation within the local density approximation (LDA) functional [34,35]. Norm-conserving pseudopotentials [51] have been employed with the following orbitals considered as the valence states: $5s$, $5p$, and $6s$ for Ba $4s$, $4p$, $4d$, and $5s$ for Zr, and $2s$ and $2p$ for O. The energy cutoff for the expansion of the electronic wave functions has been fixed to 45 Ha and we used a $6 \times 6 \times 6$ k-point mesh for the Brillouin zone sampling. Phonon calculations have been performed on the fully relaxed cubic and AFD structures. In particular, structural optimization to find the equilibrium configuration of the ions and lattice parameters were performed using the Broyden-Fletcher-Goldfarb-Shanno minimization (BFGS); the maximal absolute force tolerance was fixed to 10^{-5} Ha/Bohr. A negative and isotropic target stress tensor was imposed during the geometry optimization to simulate the hydrostatic pressure effect. The phonon dispersion curves for

the 5-atom cubic cell of BaZrO₃ have been obtained through Fourier-based interpolation of the dynamical matrices, as implemented in the postprocessing tool related to the ABINIT package, ANADDB [49]. Direct computation of the full dynamical matrices was performed via DFPT at the Γ , X , M , R , and the Λ (halfway from Γ to R) points of the simple cubic Brillouin zone. For the comparison with the Raman data and analysis, phonon frequencies of the AFD structures have been calculated at the Γ point. Symmetry analysis of the phonon modes was done with the help of programs from the Bilbao crystallographic server [52,53] and the ISOTROPY software suite [54].

III. RESULTS AND DISCUSSION

A. Raman spectroscopy

The Raman spectra collected under pressure up to 21 GPa are shown in Figs. 1(a) and 1(c). At ambient pressure, in the cubic phase, BaZrO₃ has no Raman active phonon modes but nonetheless exhibits an intense Raman spectrum consisting in broad second-order bands as described in Ref. [30] and references therein. Upon increasing pressure, this second-order spectrum weakens. At 10 GPa, some additional sharp peaks emerge, revealing the activation of Raman modes due to the structural phase transition. In addition, a soft mode comes into view from lower frequencies with high intensity [Fig. 1(a)], followed by a second soft mode at lower wavenumbers only visible when measuring very close to the quasielastic line [Fig. 1(c)]. The frequency evolution under pressure of the Raman peaks, fitted with Lorentzian functions, is shown in Figs. 1(b) and 1(d). The hard modes exhibit slopes ranging from $0.9 \text{ cm}^{-1} \text{ GPa}^{-1}$ to $2 \text{ cm}^{-1} \text{ GPa}^{-1}$ [see Figs. 1(b) and 1(d)].

An assignment of the modes appearing at the transition can be done with the help of the analysis detailed in our previous

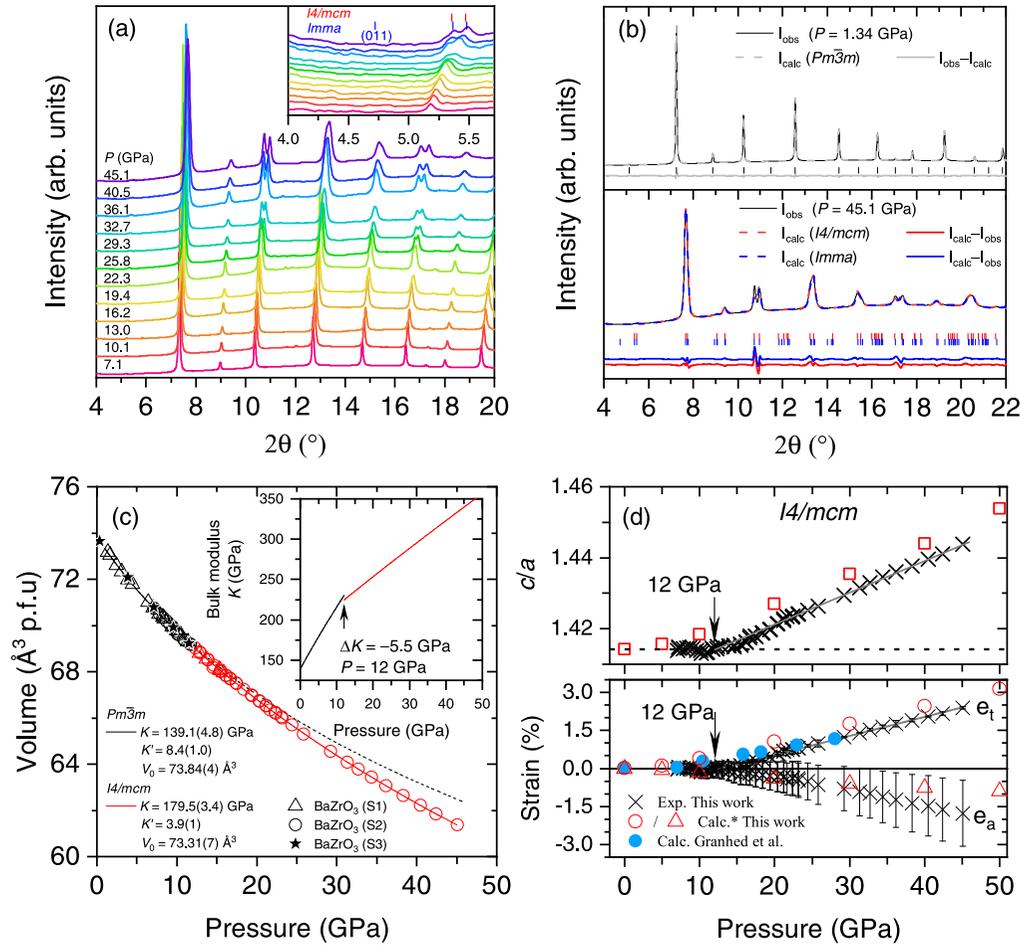


FIG. 2. (a) Diffraction patterns of sample S2. Inset: Zoom of the low-angle region. Low-angle fitted peak positions at 45.1 GPa are included as red and blue ticks corresponding to *I4/mcm* and *Imma* structures, respectively. Note that no diffraction peak is visible where the reflection (011) of an *Imma* phase would be expected. (b) Top: Rietveld fit (grey line) to a diffraction pattern (black line) acquired at 1.34 GPa. Bottom: Rietveld fit assuming tetragonal and orthorhombic phases (red and blue dashed lines, respectively) to a pattern acquired at a pressure of 45.1 GPa. Fitted positions for all reflections are included at the bottom as colored ticks. Both panels use the same intensity scale and residuals of the *Imma* fit are vertically shifted for a better comparison. (c) Pressure dependence of the volume per formula unit ABO₃ for the cubic and tetragonal phases (black and red, respectively). Lines are fits to a BM EoS (for the cubic phase a high-pressure extrapolation is included as a black-dashed line). Inset: Bulk modulus as a function of pressure. (d) Top: Tetragonality (c/a ratio) as a function of pressure as obtained from the fitted x-ray diffraction (XRD) patterns of sample S2 (crosses) and adapted DFT calculations (squares). Bottom: Volume (e_a) and tetragonal (e_t) spontaneous strains as obtained from the experiments (crosses), our calculations (open symbols) and calculations published by Granhed *et al.* [25]. Our reported DFT-values include a pressure shift (+10 GPa) as discussed in the text; this is not the case for values taken from Ref. [25] calculated through the hybrid Heyd-Scuseria-Ernzerhof (HSE) functional, which provides a slight larger volume than the one obtained from the WC-functional employed here (cf. Table I in Ref. [25]).

paper [30] and partially reproduced in Table I. The mode that appears at 124 cm^{-1} is assigned to Ba motion and the mode appearing at 384 cm^{-1} to oxygen octahedra shearing modes, while the soft modes are known to correspond to tilts of the octahedra. Only the totally symmetric soft mode can be followed with accuracy, the lower energy soft mode [visible in Fig. 1(c) and fitted in Fig. 1(d)] gets lost in the quasielastic line at low pressures. Note that the hard mode appearing at 124 cm^{-1} (at 140 cm^{-1} at 20 GPa) has an asymmetric profile at higher pressures, which may indicate the presence of two close overlapping bands.

Our single crystal spectra are overall in agreement with the single crystal spectra from Ref. [31] and the powder spectra from Ref. [14], but also differ in some aspects. The major

difference is that an intense peak appears in the powder spectrum at the phase transition and reaches 680 cm^{-1} at 20 GPa. On our single crystal spectra, we do observe a feature that becomes visible at 634 cm^{-1} around 11–12 GPa and hardens up to 670 cm^{-1} at 19 GPa [Fig. 1(a)], but with significantly less intensity. The very same mode does appear very clearly in the single crystal study by Gim *et al.* [31]. This difference in intensity between different experiments presumably arises from orientation effects due to selection rules, as will be discussed in the following.

Next, we focus on the identification of the symmetry of the high-pressure phase. In Ref. [14], the first high-pressure phase was identified as $R\bar{3}c$ by analogy with the Raman spectrum of Ba_{0.9}Ce_{0.1}ZrO₃, and because the Raman modes appeared

TABLE I. Wavenumbers (in cm^{-1}) at 20 GPa of the Raman-active phonon modes calculated through DFPT for the three tilted phases considered here as well as values measured experimentally. The full table including Raman inactive modes is included in the Supplemental Material (SM) [66]. The modes are grouped together based on the mode they originate from in the cubic phase.

Cubic ($Pm\bar{3}m - 221$)	Vibrational pattern	Tetragonal ($I4/mcm - 140$)		Orthorhombic ($Imma - 74$)		Rhombohedral ($R\bar{3}c - 167$)		Experimental frequencies
R_4^+	Oxygen octahedra rotations	76	E_g	39	B_{2g}	23	E_g	48
		189	A_{1g}	69	B_{1g}			
R_5^+	Antiparallel Ba motion	138	E_g	182	A_g	177	A_{1g}	115
		145	E_g	135	B_{2g}	142	E_g	
R_5^+	Oxygen octahedra shearing modes	374	E_g	134	B_{3g}			137
		145	B_{2g}	139	A_g			
R_5^+	Oxygen octahedra shearing modes	374	E_g	363	B_{2g}	383	E_g	
		385	B_{2g}	373	B_{3g}			394
R_3^+	Jahn-Teller-like distortions of oxygen octahedra	385	B_{2g}	390	A_g			
		665	B_{1g}	662	B_{1g}	665	E_g	672
R_1^+	Oxygen octahedra breathing mode			896	B_{3g}			

first as single peaks. It was then proposed that a second phase transition occurs at higher pressures based on the splitting of the hard Raman modes located around 150 and 380 cm^{-1} . The same reasoning was made in Ref. [31], but the second high-pressure phase was identified as tetragonal instead. As stated in our previous study, identifying the high-pressure structure from Raman spectra may not be as easy as anticipated. This is because, even though the number of Raman active modes is theoretically different, they originate in all cases from the same degenerate modes at the R point of the cubic phase, so that differences between the tetragonal, orthorhombic, and rhombohedral structures are only revealed by peak splittings that scale with the very small distortion of the unit cell, and mode polarizations that are difficult to appreciate without performing polarized experiments on crystals with known orientation. In such a context, identifying the correct number of Raman active modes can be troublesome.

In order to properly identify the high-pressure phase, we provide in Table I the wavenumbers of the Raman active modes computed by DFPT at 20 GPa for the three considered phases with antiphase tilts (tetragonal $I4/mcm$, orthorhombic $Imma$, and rhombohedral $R\bar{3}c$) together with our experimental Raman values at 20 GPa. From this table, we make the following observations. First, the orthorhombic phase is markedly different from the two others mainly by three criteria: (i) splitting of the soft mode into three modes instead of two, (ii) large splitting of the second hard mode with a totally symmetric mode given here at 390 cm^{-1} , and (iii) Raman activation of the octahedra breathing mode at 896 cm^{-1} . Octahedra breathing modes in perovskites usually have strong intensities when symmetry allowed, but here they unfortunately overlap with very strong bands from the pressure transmitting medium and nothing can be said about this mode neither in our paper nor in Refs. [14,31]. On the other hand, none of these three Raman studies shows a peak splitting of the second hard mode. It appears very unlikely that such a splitting could occur without being detected, since the modes all originate from the same atomic displacement patterns and would have *a priori* comparable Raman intensities, especially in a powder average. We therefore exclude here the possibility of an or-

thorhombic phase. Distinguishing between the rhombohedral and the tetragonal phase is more delicate. They differ only in mode symmetries and with very moderate peak splitting (smaller than 10 cm^{-1}), which can be easily overlooked given the observed peak widths. A detailed examination of mode symmetries strongly supports polarization effects as the origin for the differences observed between the powder spectra from Ref. [14] and our single crystal spectra. Hard modes split into modes of different symmetries, which is more easily observed with the orientational average of a powder than on single crystals. Interestingly, the hard mode appearing at 672 cm^{-1} is predicted to have a B_{1g} symmetry in the tetragonal phase, which makes it particularly difficult to observe on single crystal measurements if the tetragonal c axis is in the plane of the DAC. In contrast, it has the same symmetry (E_g) as the other hard modes in the rhombohedral phase, and no particular reason to be less visible. Therefore, comparison between our single crystal and the powder spectra from Ref. [14] points towards the tetragonal phase as the most probable. This assignment becomes in turn perfectly compatible with the peak splitting reported at higher pressures: At low pressures close to the transitions, peak splitting is very weak and these doublets $E_g \oplus B_{2g}$ appear as a single peak; the splitting is resolved only at higher pressures when the distortion become sizable (at 30 GPa in Chemarin *et al.* [14] and 19.2 GPa in Gim *et al.*). With this reasoning, we do agree with Gim *et al.* [31] on their assignment of the high-pressure phase to the tetragonal $I4/mcm$ variant, but disagree on the presence of an intermediate rhombohedral phase, which is not necessary to explain the observations, and also would lead to first-order rhombohedral-to-tetragonal transition that is not compatible with the smooth evolutions observed in all studies. In summary, Raman spectroscopy supports the hypothesis of a single phase transition towards a tetragonal $I4/mcm$ phase and no further transition up to the maximum pressure of 42 GPa reached by Gim *et al.*

The exact same discussion can be made about SrTiO_3 and the conflicts on its Raman spectrum, with the additional simplicity that literature agrees on a single tetragonal phase. The splittings of the hard modes at 185 and 470 cm^{-1} reported

in Ref. [11] and interpreted as a possible phase transition towards an orthorhombic phase remain in fact compatible with a tetragonal symmetry. Also, the mode around 600 cm^{-1} is seen less clearly on a single crystal [12] than on powder [13]. Even though this mode is not clearly assigned in those past studies, it is now clear that it follows the behavior and polarization effects described above and in Table I. We also note that the Raman signatures (gradual vanishing of the second-order Raman bands, emergence of the E_g soft mode) are essentially identical to the signatures reported for SrTiO_3 in Ref. [12] and are all compatible with the tetragonal structure. Altogether, SrTiO_3 and BaZrO_3 appear as perfectly isostructural in their high-pressure behavior from the Raman point of view.

B. X-ray diffraction

Diffraction patterns measured for sample S2 are shown in Fig. 2(a) up to a pressure of 45.1 GPa. A splitting of the peaks corresponding to the pseudocubic $(200)_{pc}$ and $(310)_{pc}$ reflections around 10.6° and 17° respectively is particularly visible from a pressure of 25.8 GPa. Other pseudocubic reflections show broadening with potentially unresolved peak splitting. The rhombohedral structure is not compatible with the splitting of the $(200)_{pc}$ reflection and can therefore be safely ruled out. (See also Fig. S5 in the Supplemental Material (SM) [66] for more details). On the other hand, distinguishing between the tetragonal and orthorhombic structures may not be trivial if the orthorhombic distortion is small and the metric quasitetragonal. One possible criterion is to check the low-angle region where a $(011)_o$ reflection is expected for the orthorhombic phase only as a result of the off-centering of the Ba cation. The inset in Fig. 2(a) shows an enlarged view of this region where no reflection can be seen even at the highest pressures. From Rietveld simulations for the *Imma* structure and with our level of noise, this indicates that Ba displacement cannot be larger than ≈ 0.007 , a value that is small but not unrealistically small as we realized when comparing to values found in our DFT results. This criterion alone is therefore not sufficient to conclusively rule out the orthorhombic phase.

Another criterion is to examine the values of the (pseudo)tetragonality, based on the following observation. Both the *I4/mcm* and the *Imma* unit cells are similar, with one lattice vector that is doubled with respect to a primitive cubic lattice vector. In the *I4/mcm* phase, the octahedra rotation axis is along this doubled axis, which is then elongated while the perpendicular axes shrink, resulting in a tetragonality c/a larger than $\sqrt{2}$. In contrast, in the *Imma* structure, the rotation axis is perpendicular to the doubled axis (b in the standard setting) so that this axis shrinks and the resulting pseudotetragonality is lower than $\sqrt{2}$. This is verified experimentally notably in the *Imma* \rightarrow *I4/mcm* transition in SrZrO_3 [4] as well as in other *Imma* perovskites [55,56]. Here, the c/a ratio is most definitely larger than $\sqrt{2}$. We can therefore discard the orthorhombic phase and retain the tetragonal *I4/mcm* as the only high-pressure phase.

Rietveld refinement of the patterns were performed over the full pressure range. As illustrated in the top panel of Fig. 2(b), the diffraction pattern of S1 can be perfectly fitted at 1.34 GPa in the cubic phase. Refinements are less satisfactory

at higher pressures, as illustrated with the pattern at 45.1 GPa shown in the bottom panel of Fig. 2(b). This is expected due to a general degradation of the pattern quality at high pressure following intergrain stress, loss of hydrostaticity, etc. Unfortunately, the quality of the pattern was not sufficient to determine reliably the oxygen positions, it is therefore not possible to give a value for the tilt angle itself. For completeness, we also performed a Rietveld refinement of the pattern at 45.1 GPa in the orthorhombic phase [also displayed in Fig. 2(b)], only to note that it does not improve the quality of the refinement as compared to the tetragonal case (reliability factors are, for *Imma*, $R_p = 16.3$, $R_{wp} = 13.0$ and $\chi^2 = 0.58$ while slightly better values are found for *I4/mcm* $R_p = 12.8$, $R_{wp} = 12.3$, and $\chi^2 = 0.59$ due to small differences in the fitting procedure). This further validates the tetragonal assignment.

The pressure-volume relation of S1, S2, and S3 are shown in Fig. 2(c). All data were adequately fitted using the third-order Birch-Murnaghan (BM) EoS (a discussion on the choice of EoS is provided in Fig. S6 of the SM based on finite-strain analysis [57]), fitted volumes at zero pressure as well as bulk moduli and its pressure derivative are provided in the figure. For the tetragonal phase, very small deviations (i.e., volume deviations smaller than 0.1%, as shown in Fig. S7 in the SM [66]) are found from the fitted EoS up to the highest pressure, which is another indication that no other phase transition takes place and that only the *I4/mcm* structure remains stable at least up to 45.1 GPa. There is no measurable volume jump at the transition. The pressure dependence of the bulk modulus in both phases has been included as an inset to Fig. 2(d) in order to better illustrate the changes of second derivative of the Gibbs energy at the transition pressure. As it can be seen in the inset, at the transition pressure the bulk modulus exhibits a decrease of slope (due to the decreased K' for the high-pressure phase) together with a small variation ($\Delta K = -5.5\text{ GPa}$), which is consistent with a second-order transition.

From the Rietveld refinements we extracted the lattice parameters for both phases for samples S1 and S2. The tetragonality c/a is plotted versus pressure in Fig. 2(d). We adopt here the symmetry-adapted spontaneous strains defined as $e_a = (2e_1 + e_3)$ (volume strain) and $e_t = 2(e_3 - e_2)/\sqrt{3}$ (tetragonal strain), where e_1 and e_3 are the spontaneous strain components of the tetragonal phase with respect the cubic phase, $e_1 = e_2 = (a_{pc} - a_0)/a_0$ and $e_3 = (c_{pc} - a_0)/a_0$ [58]. We consider $a_0 = V^{1/3}$, where V is a cubic volume extrapolated from the EoS of the cubic phase whereas the pseudocubic lattice constants were calculated from the tetragonal lattice constants, $a_{pc} = a_t/\sqrt{2}$ and $c_{pc} = c_t/2$. As it can be seen in the lower panel of Fig. 2(d) the tetragonal strain e_t is positive and linear with pressure with a slope of $7.5 \times 10^{-4}\text{ GPa}^{-1}$. The volume strain e_a is negative and exhibits comparatively large errors due to the increasing uncertainty of the extrapolated EoS at higher pressure. Good agreement is found with respect to the DFT calculations, as shown in Fig. 2(d). Noteworthy, both e_a and e_t values are very similar to those reported for SrTiO_3 [12].

Overall, the combination of XRD and Raman spectroscopy data confirms that BaZrO_3 undergoes a phase transition from cubic to tetragonal but also show that no other transition happens up to the maximum experimental pressure investigated of 45.1 GPa.

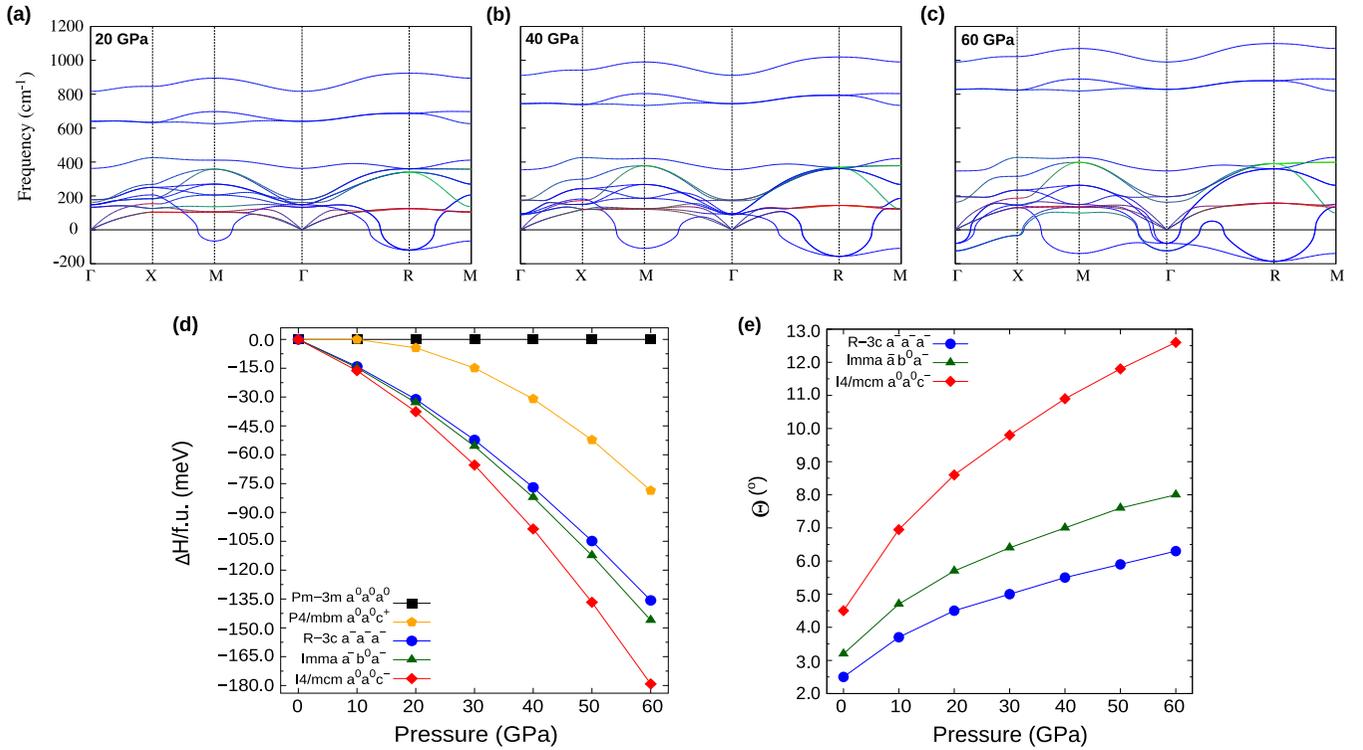


FIG. 3. [(a),(b),(c)] DFPT phonon dispersion curves in the cubic phase for 20, 40, and 60 GPa applied hydrostatic pressure (corresponding lattice parameters a are 4.049 Å, 3.955 Å, and 3.883 Å, respectively). Red, green, and blue phononic branches involve mostly Ba, Zr, and O vibrations, respectively. (d) Computed enthalpies (H), with respect to the cubic phase taken as reference at each pressure. (e) Evolution of the oxygen octahedra rotation angle Θ (in degree, °) with hydrostatic external pressure for different rotation patterns.

The value of the transition pressure varies slightly depending on the method used. From the powder diffraction data, it can be determined by an extrapolation of the tetragonality to its ideal cubic value of $\sqrt{2}$ (considering c_t/a_t), or of the tetragonal strain to zero, which gives a value of 12 GPa. This is in line with the emergence of superstructure reflections observed in the single crystal diffraction patterns only slightly above the noise level at 11.2 GPa and much more conclusively at 12.5 GPa (Fig. S4 in the SM [66]). In the Raman data, the first-order peaks appear at 10 GPa, which is consistent with the ≈ 11 GPa measured from Raman spectroscopy in Ref. [14]. The slightly lower value found by Raman spectroscopy as compared to XRD is consistent with the smaller coherence length of this technique and its sensitivity to small structural changes. We note that our values are significantly lower than the 17.2 GPa given in Ref. [15] where it was determined only from the observation of the peak splitting in a powder XRD pattern. In the following, we chose to retain the value of 10 GPa measured by Raman spectroscopy.

C. First-principle calculations

The propensity of BaZrO₃ to undergo a pressure-induced structural phase transition can be inferred from the inspection of phonon frequencies calculated in the compressed cubic reference structure, as well as from the DFPT-calculated phonon dispersion curves shown in Figs. 3(a)–3(c) (and Fig. S1 in the SM [66]; cf. also Ref. [34]). As it can be seen in the figure, an increase of external hydrostatic pressure (P) results in enhanced instabilities of the high-symmetry cubic phase

arising from oxygen motion. In fact, increasing P enhances the AFD instabilities associated with the cooperative rotations of the oxygen octahedral cage [Fig. 3(d)]. For a wide range of pressure, up to 40 GPa, the main driving instability is the one at R , accompanied by that at M , which is its continuation in the phonon spectrum. Higher pressure destabilize the system further, making polar and antipolar instabilities appearing at the Γ and X points, respectively.

It should be noted that, from our DFT calculations, the cubic phase is already unstable at 0 GPa [22–25,30], but can be made stable with a negative pressure of about -10 GPa (Fig. S1 in the SM [66]). Such behavior can be ascribed to known effects related to the used exchange-correlation functional in the calculations [59]. Besides, first-principles simulations do not account for temperature effects, which can strongly affect the critical pressure (P_c) at which the phase transition occurs, shifting it towards a higher value. For example, the phase diagram of SrTiO₃ exhibits a P - T slope of 0.054 GPa K^{-1} , which causes a shift in transition pressure of $\Delta P_c = 16.2 \text{ GPa}$ between 0 and 300 K [12]. Here, a shift of $+10 \text{ GPa}$ has been added to the DFT data to match the transition point, when comparing the calculated c/a ratio and spontaneous strains with the experimental ones. This is here appropriate since we are interested in trends and the energetics of the different phases rather than in the accurate and absolute values of transition pressures. On a related note, the thermal expansion is not strong here. In fact, the relaxed lattice parameter (by DFT) of the cubic structure at 0 K – 0 GPa ($\approx 4.184 \text{ \AA}$) perfectly reproduces the low-temperature ($T = 4.2 \text{ K}$) experimental value ($\approx 4.188 \text{ \AA}$), but is also very similar

to that measured at room temperature ($\simeq 4.193\text{\AA}$) [60]. A good agreement is also found for the volume and lattice parameters of the AFD structures upon external pressure, with respect to those experimentally estimated via the Rietveld refinements of the XRD data (Figs. S3a and S3b in the SM [66]).

Next, we compared the energetics associated to the different possible AFD phases in order to evaluate the possible phase transitions. An analysis of the computed and measured Raman active modes for candidate structures was addressed in Sec. III A. In Fig. 3(d), we show the evolution of the thermodynamic potential energy, the enthalpy $H = U + PV$, where U is the DFT-calculated internal energy of the system associated with the relaxed volume V of the AFD-structures under the constraint of given hydrostatic stress (σ_{ij}), corresponding to the external hydrostatic pressure P . We see that there is a continuous increase of the energy gain as a function of pressure for the considered AFD structures with respect to the reference cubic. In particular, the tetragonal $I4/mcm$ phase, to which corresponds the biggest oxygen rotation angle Θ [Fig. 3(e)], is the lowest energy configuration at all studied pressures. Noteworthy, the in-phase ($a^0a^0c^+$) rotation pattern, which define the tetragonal partner $P4/mbm$, produces a much lower gain of energy. In particular, in the compressed cubic cell, it can be seen that, there is a significant imbalance between the energy gain brought by the condensation of the c^+ rotation and by the c^- ones, in favor of the latter, as shown in Fig. S3(c) in the SM [66]. This behavior is similar to what also observed in SrTiO_3 and PbTiO_3 [61,62], and thus it could be a typical feature of perovskite oxides with a tolerance factor t close to 1. Clearly, the interplay between pressure and strain relaxation stabilize the $a^0a^0c^-$ -AFD tetragonal phase with $I4/mcm$ space group at the expense of all other candidates. Moreover, the phonon spectrum of the $I4/mcm$ phase ($a^0a^0c^-$), calculated at 20 GPa, does not display any instability or any mode softening (cf. Table SI and Fig. S2 in the SM [66]), confirming that the tetragonal phase is dynamically stable and suggesting that no further phase transition is expected in the investigated pressure range.

We also considered the possibility for a transition to the ($a^-b^+a^-$) tilt system, corresponding to the very common and stable orthorhombic $Pnma$ structure. For this case, the system relaxes back to the orthorhombic $Imma$ ($a^-b^0a^-$) phase suppressing the b^+ rotation, also in line with past studies [38,63]. Therefore, this highlights that, in BaZrO_3 , the condensation of one rotational mode suppresses the partner one. It can be surprising at first sight, given how common the $Pnma$ structure is in perovskite systems and the various tilt instabilities appearing at high pressure (e.g., at 60 GPa). Nevertheless, a careful analysis of the phonon dispersion and character of the modes at the different high-symmetry k points of the cubic-BZ zone reveals crucial features. In $Pnma$ perovskites such as CaTiO_3 and SrZrO_3 at 0 GPa [5,28], it is the competition and coexistence of in-phase and antiphase rotations together with the trilinear coupling of the latter modes with the A-cation antipolar displacement, in turn also linked to the existence of an unstable polar zone-center mode in the parent cubic phase, mainly driven by the A cation, to lower the ground state energy [5,64,65]. Instabilities involving displacements of the A cation is therefore a key ingredient that is missing in BaZrO_3 at high pressure, and a fortiori at ambient pressure. At ambient pressure, cubic BaZrO_3 displays neither the unstable

in-phase tilt mode nor any polar instability. Under pressure, these modes become indeed unstable but (i) the pattern of distortion associated to the unstable Γ mode is mainly characterized by a polar motion of the zirconium atoms at the B site against the oxygen cage; (ii) the phonon branch related to the Ba displacements stay higher in energy with respect to the CaTiO_3 and SrZrO_3 cases. As a result, the $Pnma$ phase is never stabilized.

IV. CONCLUSION

In summary, we have reported a combination of Raman spectroscopy, x-ray diffraction, and first-principle calculations aimed at clarifying the high-pressure behavior of BaZrO_3 . We confirm the picture according to which BaZrO_3 undergoes a single phase transition around 10 GPa to a tetragonal $I4/mcm$ phase and retains this structure until the highest pressures investigated—here 45.1 GPa. We have reconciled this result with previous claims for the existence of a second phase transition, and in particular clarified the Raman mode assignment of the high-pressure phase. From our analysis, it turns out that a second transition to an orthorhombic phase or any other tilted phase is, in fact, not expected. Additionally, we explain why a transition to the common orthorhombic $Pnma$ phase is not expected for this perovskite system. Instead, our study draws a picture where the stability and dominance of the tetragonal phase with its single antiphase tilt is reinforced with high pressure, with a tilt angle that can reach values as large as 13° at 60 GPa according to DFT. The question of its stability limit at even higher pressures therefore remains an open question. According to the calculated phonon dispersion curves, softening of polar modes, and subsequent competition with antiferrodistortive distortions should be expected. In particular, following other recent theoretical studies, two candidate phase transitions can be considered: The first is a transition to a polar phase combining tilts and polar cation displacements; the second is a direct transition to a post-perovskite phase. We anticipate that measurements in the 100 GPa (MBar) range, possibly complemented by *in situ* heating to overcome energy barriers, will be necessary to find out which of those two possible options prevail.

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